

Gold(I) Catalyzed Enantioselective Reactions

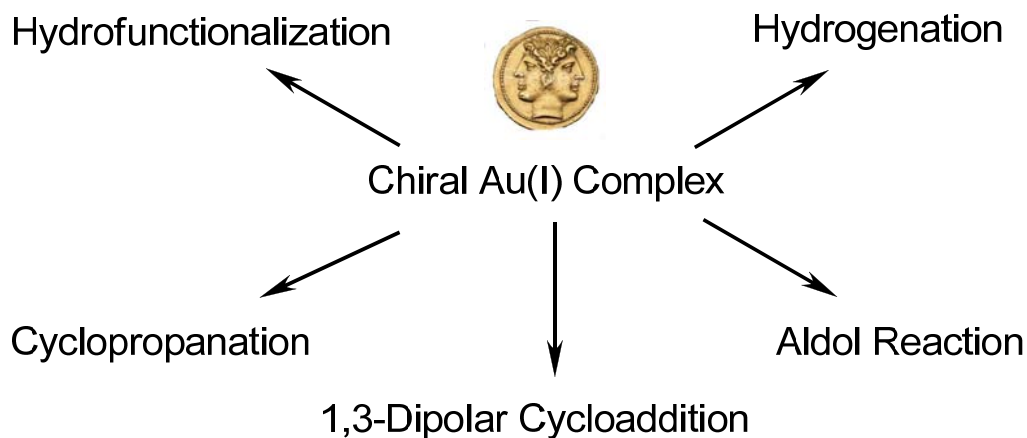
Li HUANG

Michigan State University

November 28, 2007

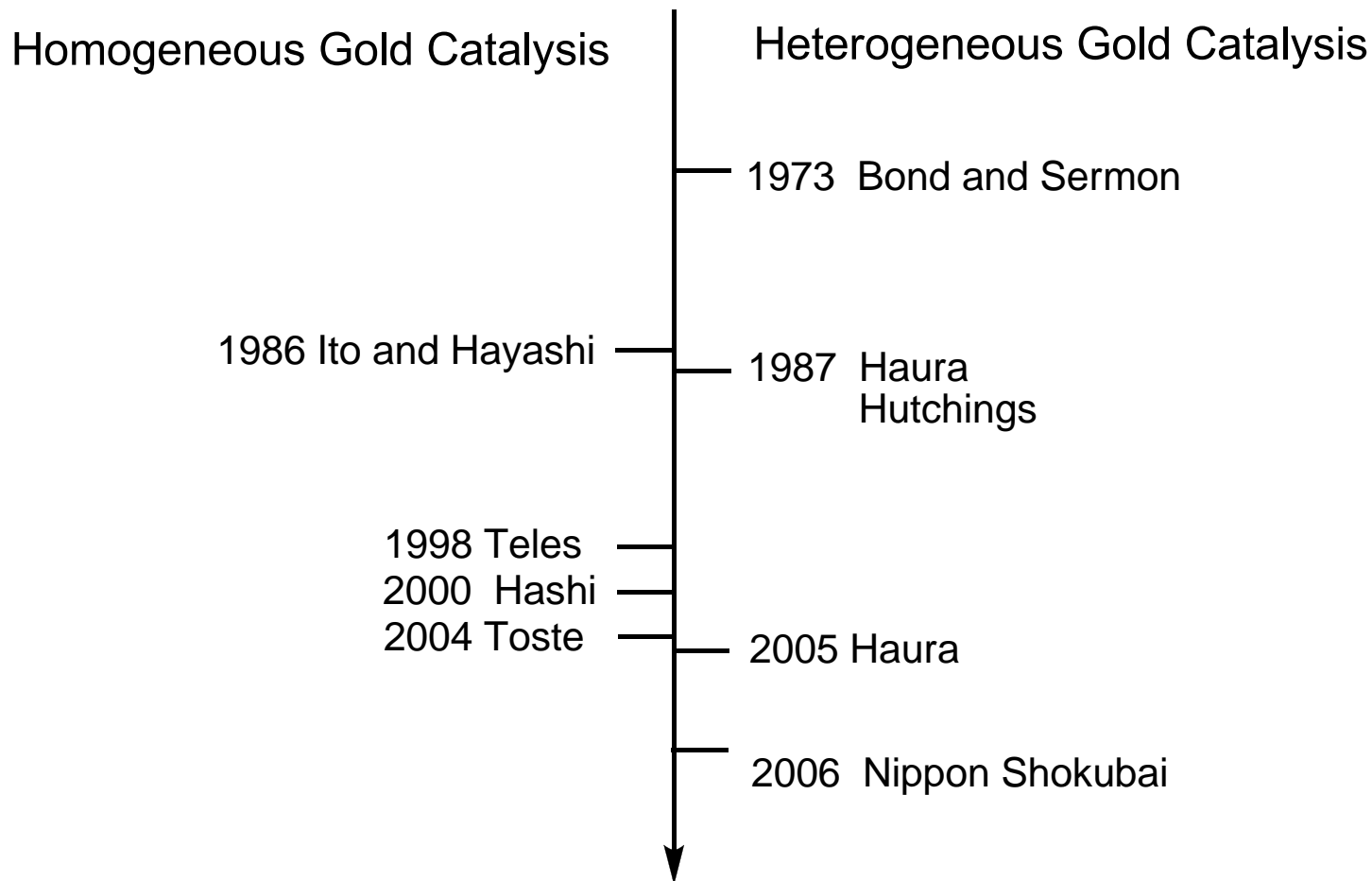
Outline

- Introduction
- Enantioselective Reactions



- Conclusions

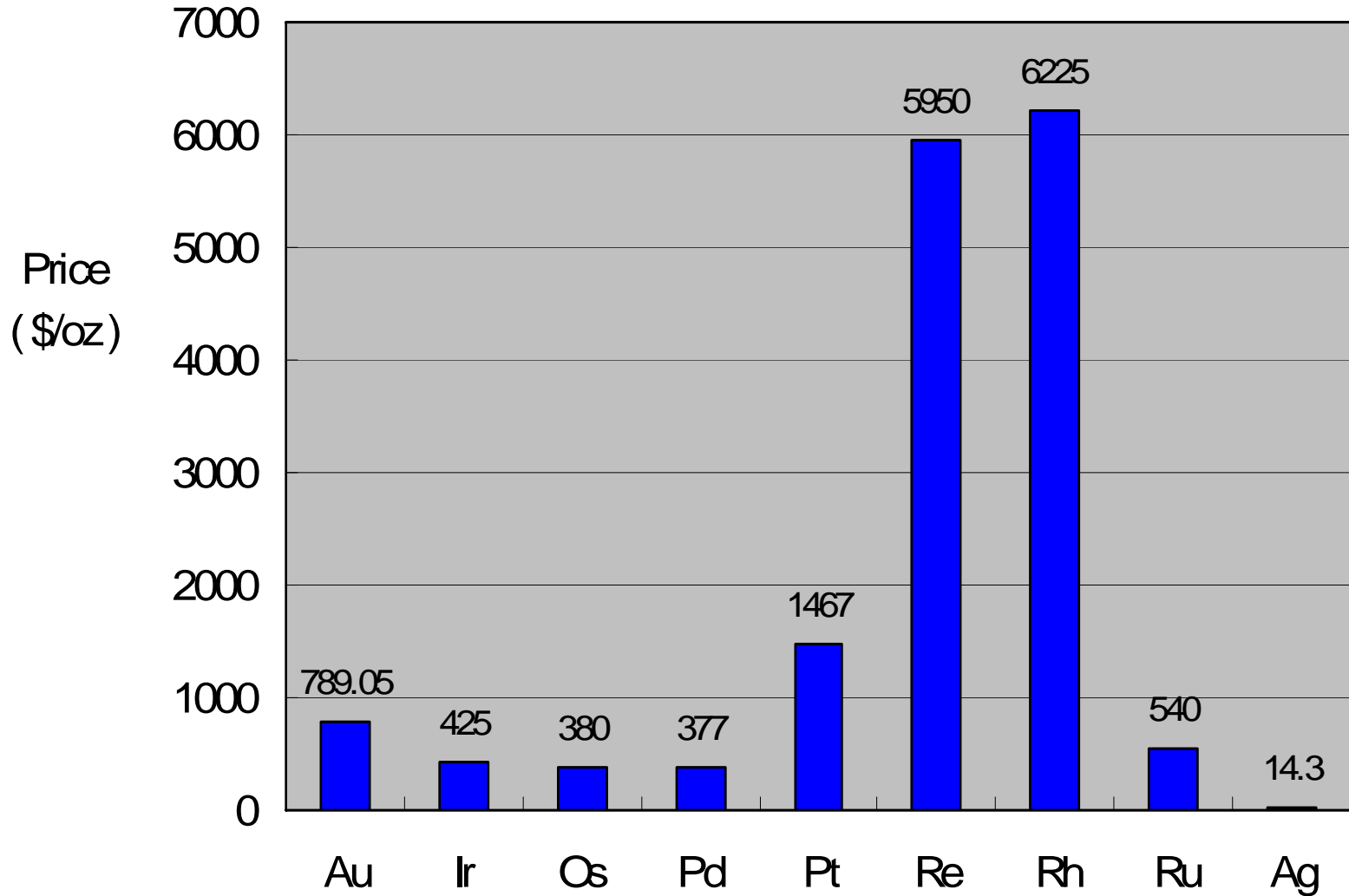
History of Gold Catalysis



Hashmi, A.S. K.; Hutchings, G. J. *Angew. Chem. Int. Ed.* **2006**, *45*, 7896.

Burks, R. *Chem. Eng. News.* **2007**, *85*, 87.

Comparison of Metal Prices



Au in Periodic Table

Periodic Table of the Elements

1	IA																O																			
1	H	IIA															2	He																		
2	3	Li	4	Be											5	B	6	C	7	N	8	O	9	F	10	Ne										
3	11	Na	12	Mg											13	Al	14	Si	15	P	16	S	17	Cl	18	Ar										
4	19	K	20	Ca	21	Sc	22	Ti	23	V	24	Cr	25	Mn	26	Fe	27	Co	28	Ni	29	Cu	30	Zn	31	Ga	32	Ge	33	As	34	Se	35	Br	36	Kr
5	37	Rb	38	Sr	39	Y	40	Zr	41	Nb	42	Mo	43	Tc	44	Ru	45	Rh	46	Pd	47	Ag	48	Cd	49	In	50	Sn	51	Sb	52	Te	53	I	54	Xe
6	55	Cs	56	Ba	57	*La	72	Hf	73	Ta	74	W	75	Re	76	Os	77	Ir	78	Pt	79	Au	80	Hg	81	Tl	82	Pb	83	Bi	84	Po	85	At	86	Rn
7	87	Fr	88	Ra	89	+Ac	104	Rf	105	Ha	106	Sg	107	Ns	108	Hs	109	Mt	110	110	111	111	112	112	113	113										

79

Au

* Lanthanide Series

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu

+ Actinide Series

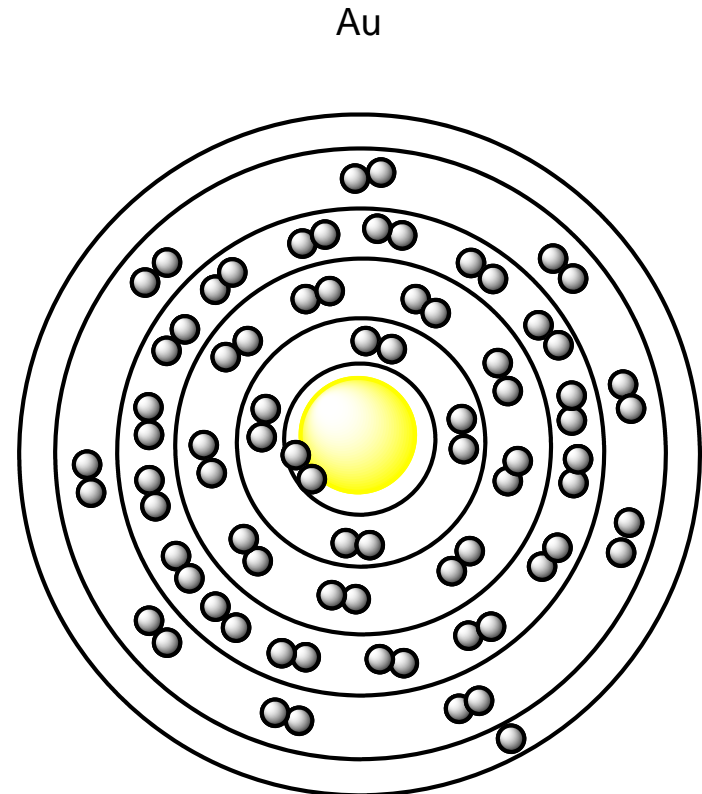
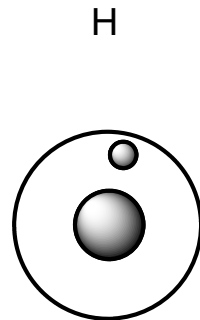
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Relativistic Effects of Au

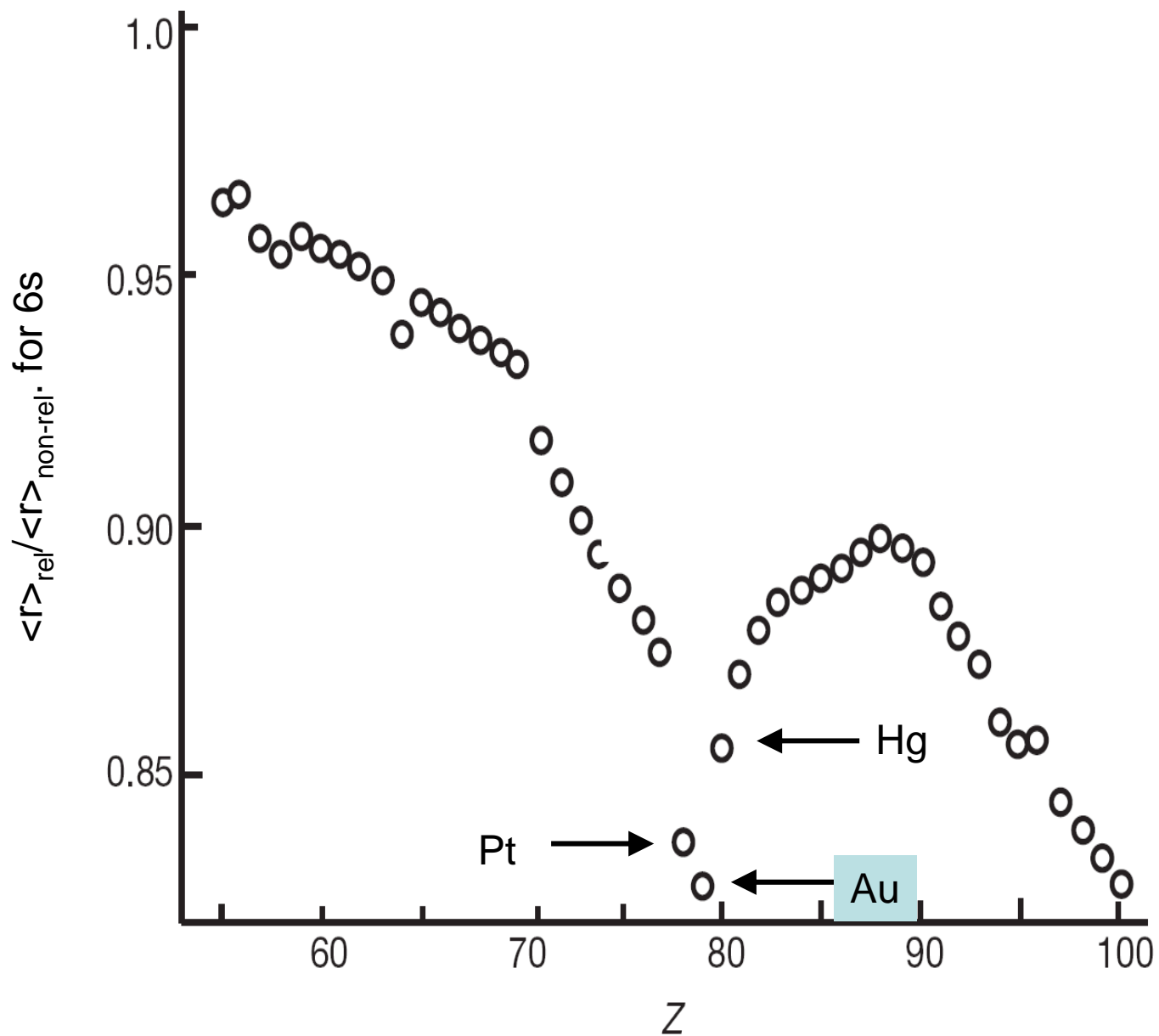
Relativistic effects

Any phenomenon resulting from the need to consider velocity as significant relative to the speed of light

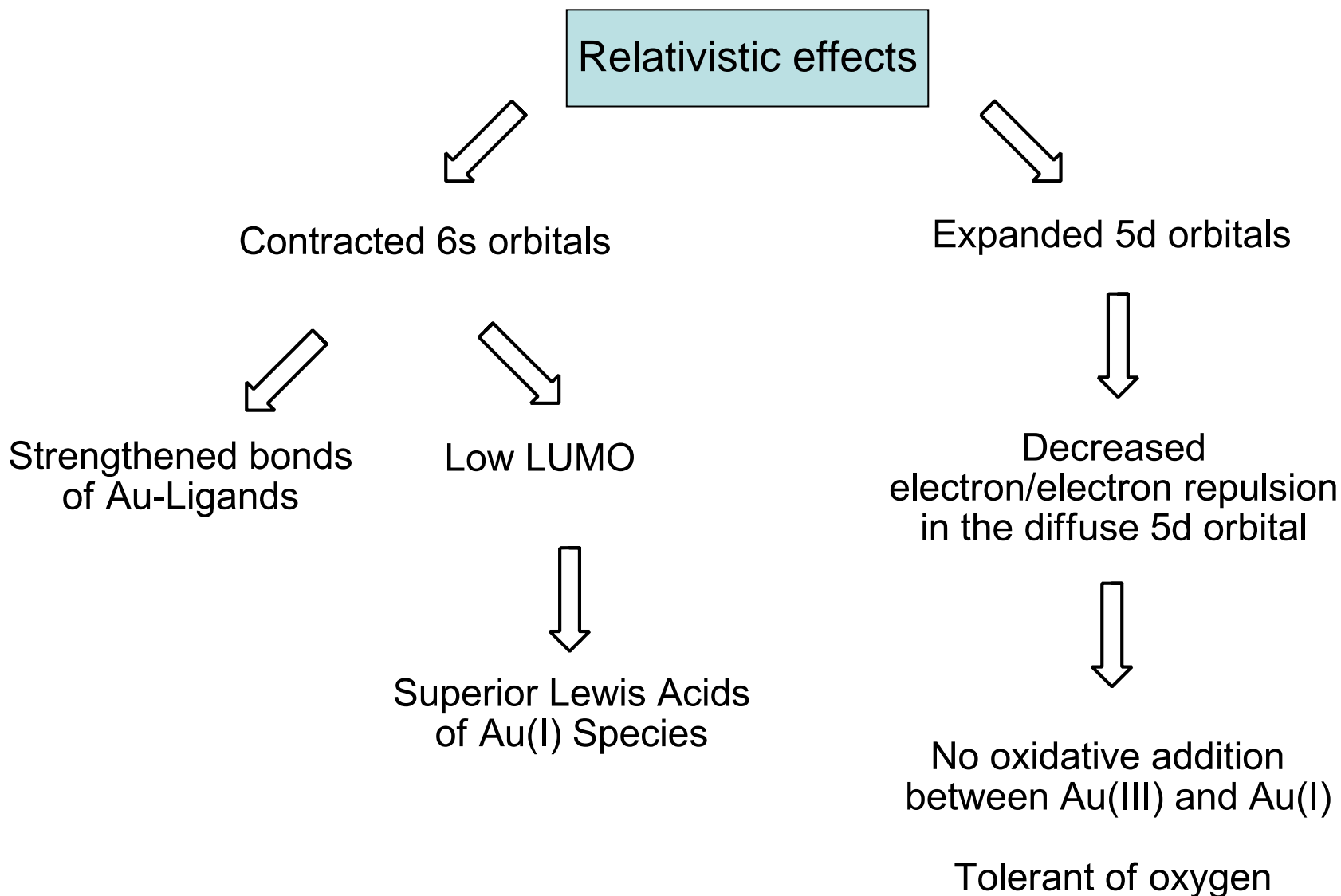
Relativistic contraction of s , p orbitals
Relativistic expansion of d , f orbitals



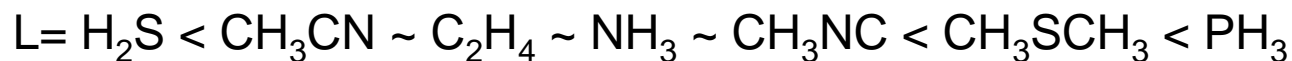
Relativistic Effects of Au



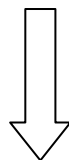
Consequences of Relativistic Effects



Lewis Acidity of Au(I) Species



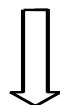
'Soft' Lewis Acid



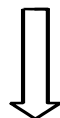
Preferentially Activating π System

Challenge in Gold(I) Catalyzed Enantioselective Reactions

Au(I) predominately adopts a linear, bicoordinate geometry.

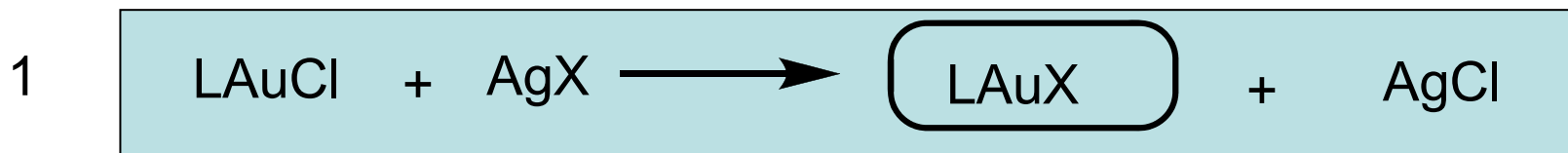


The chiral components would be distant from the substrates.

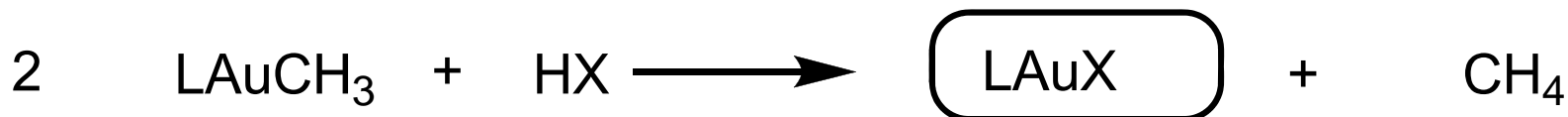


It is hard to control the enantioselectivity.

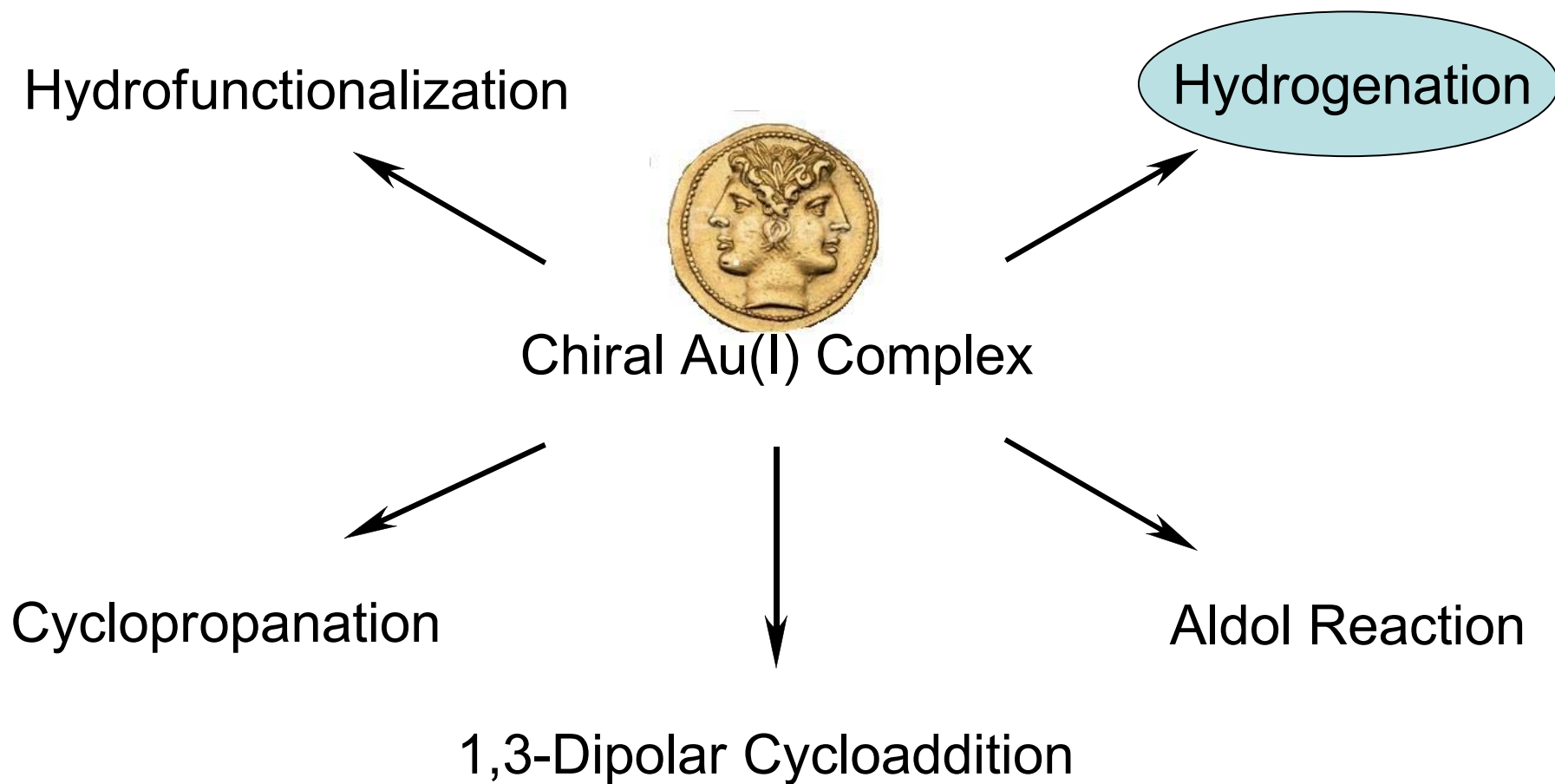
Methods to Generate Cationic Au Catalyst



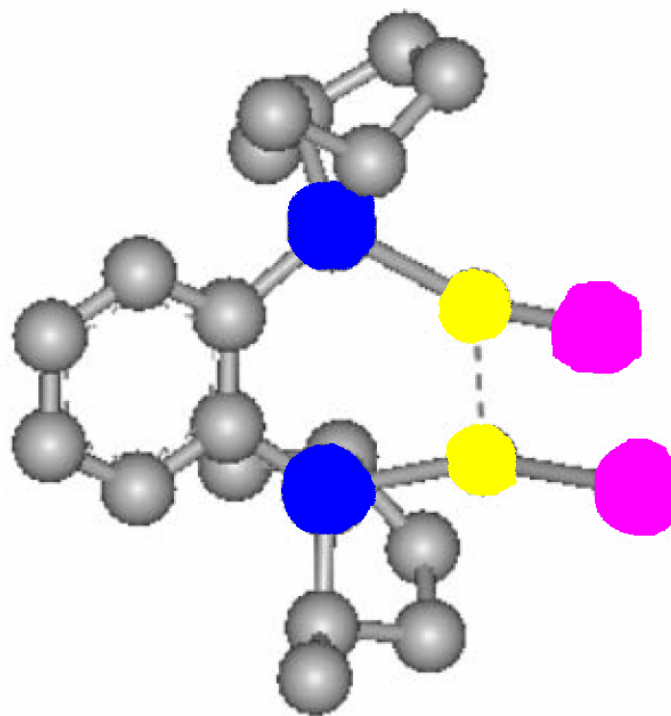
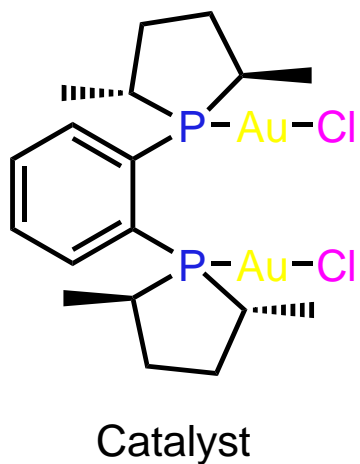
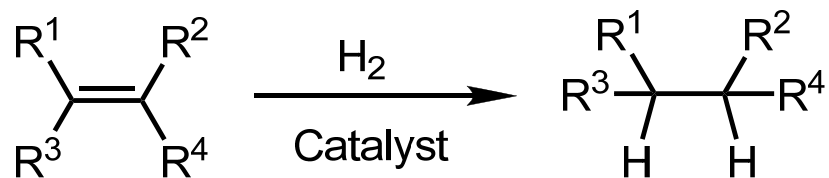
Active species



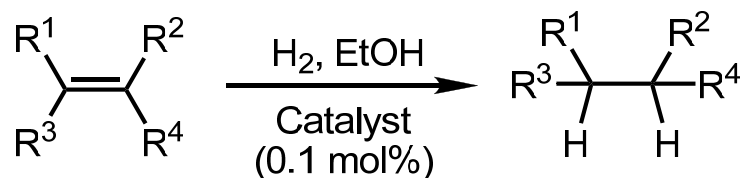
Enantioselective Reactions

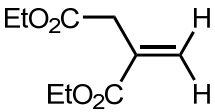
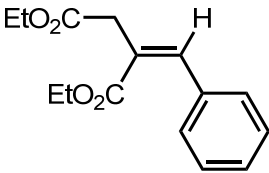
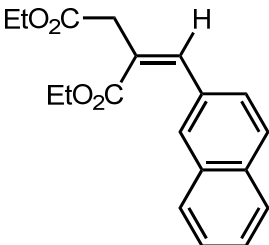
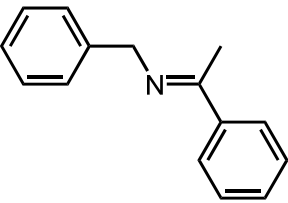


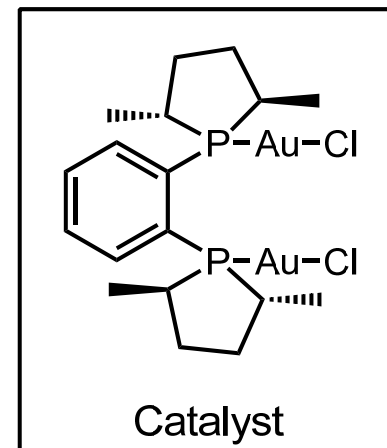
Hydrogenation Reactions



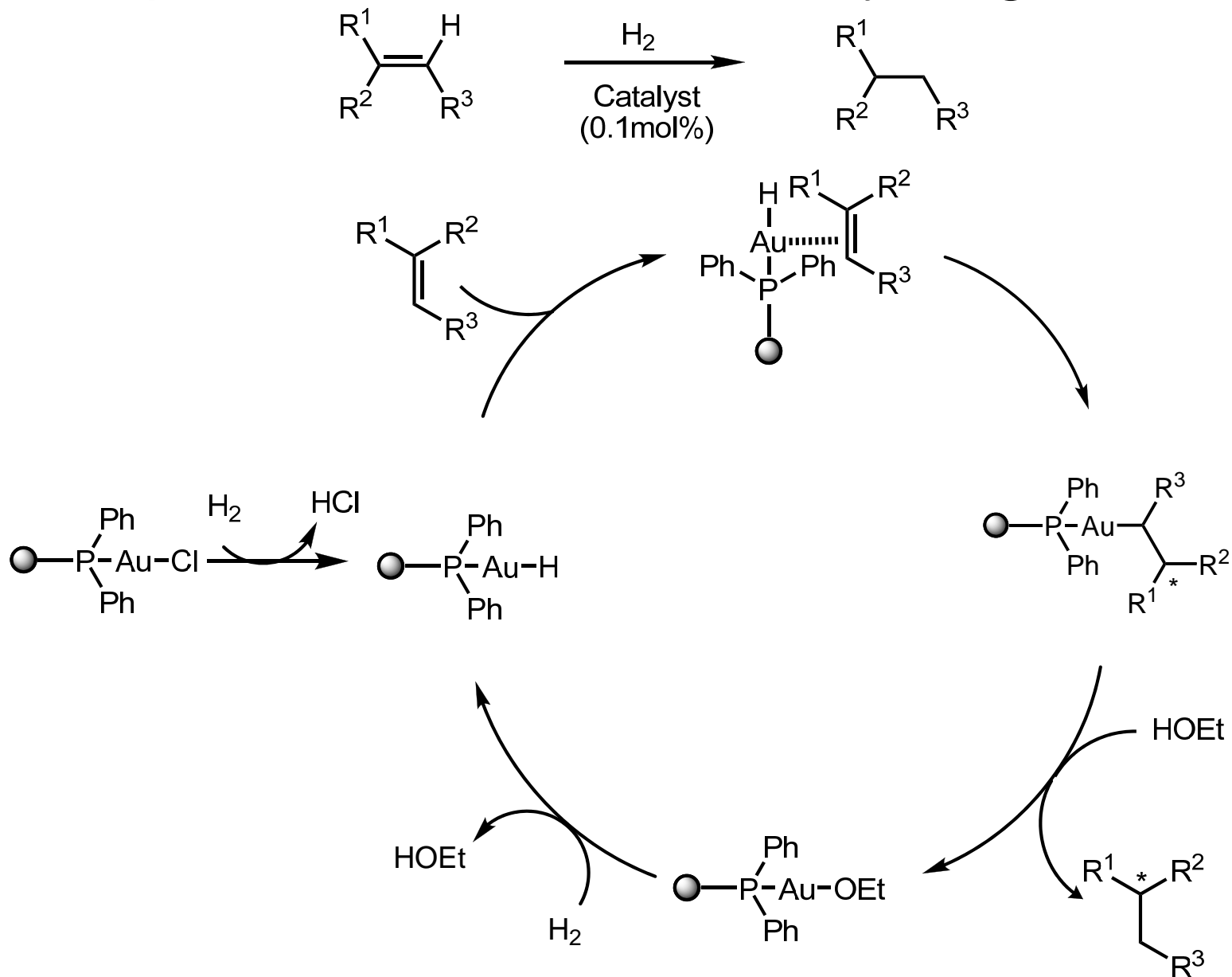
Hydrogenation Reactions



	Substrate	time (min)	ee (%)
1		20	20
2		50	80
3		150	95
4		~1920	75



Proposed Mechanism for Hydrogenation



Enantioselective Reactions

Hydrofunctionalization

Hydrogenation



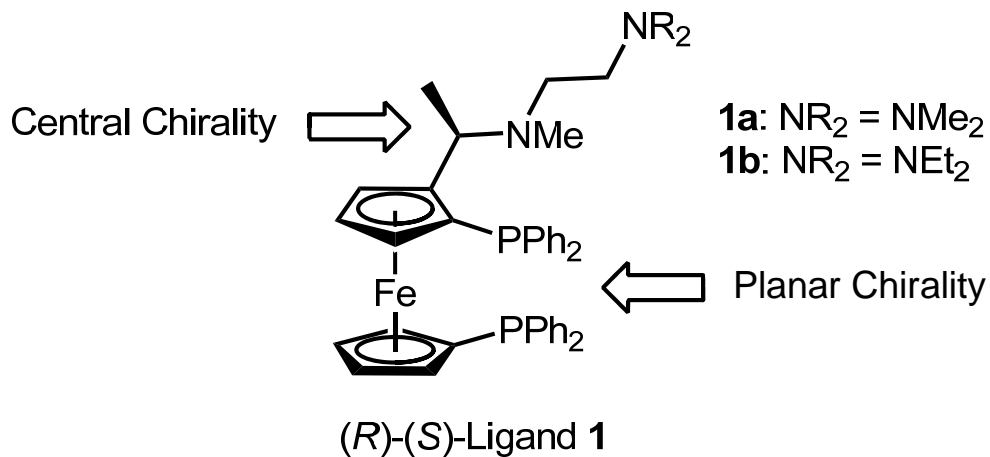
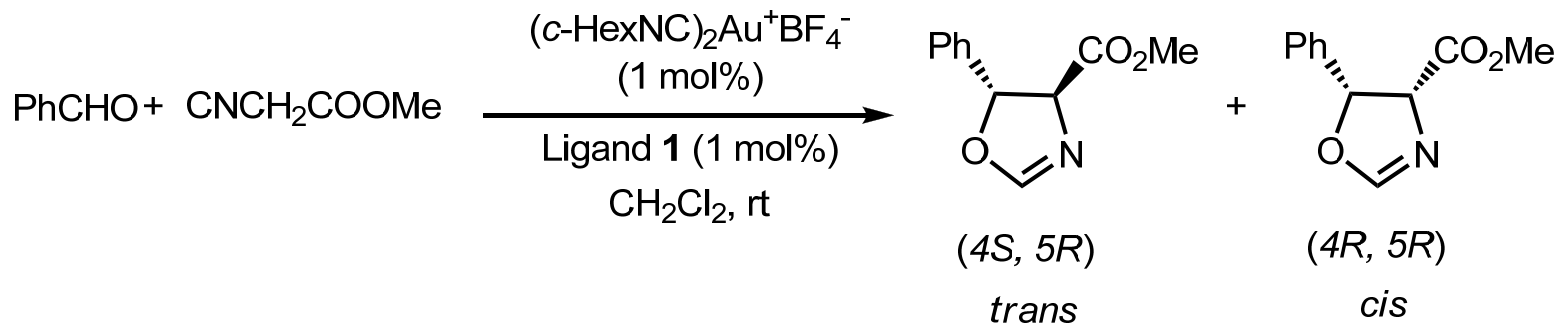
Chiral Au(I) Complex

Cyclopropanation

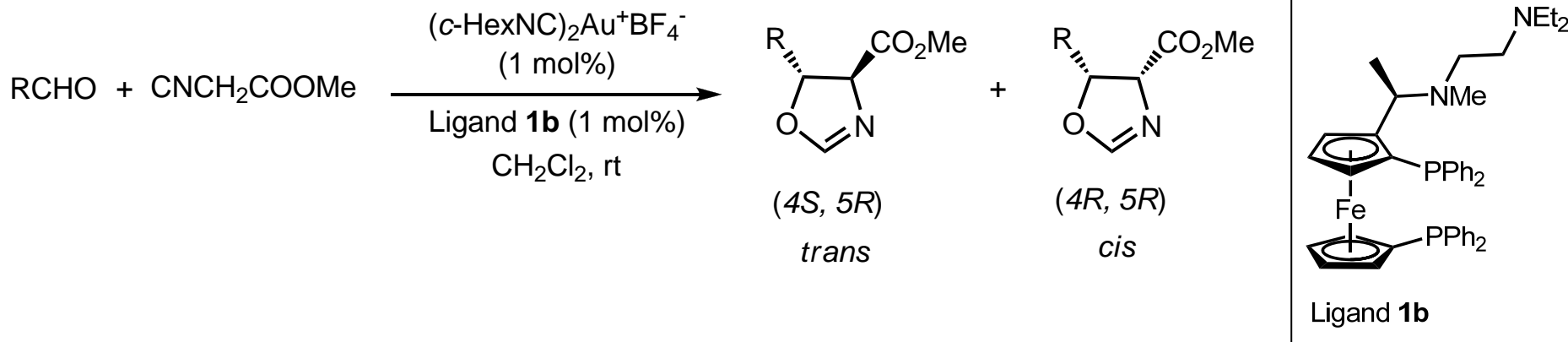
Aldol Reaction

1,3-Dipolar Cycloaddition

Catalytic Asymmetric Aldol Reaction

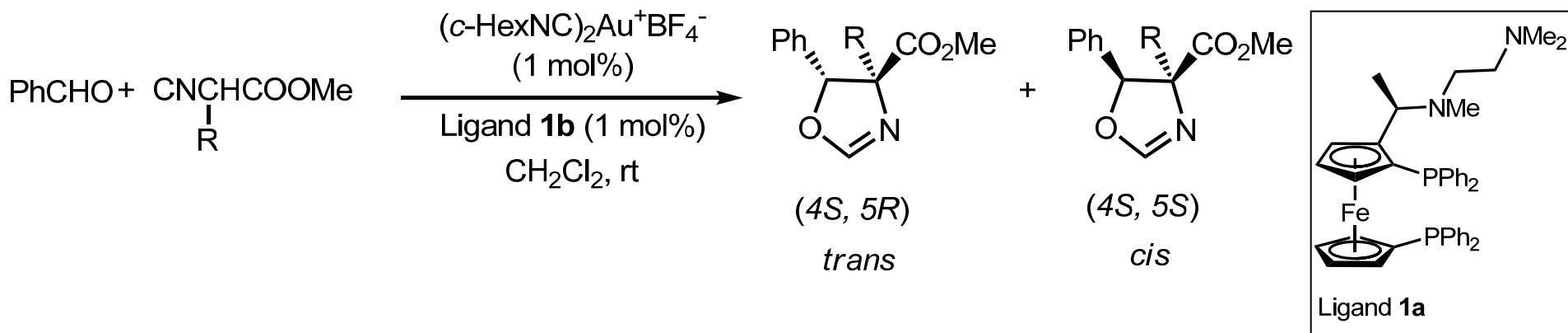


Aldol Reaction of Different Aldehydes



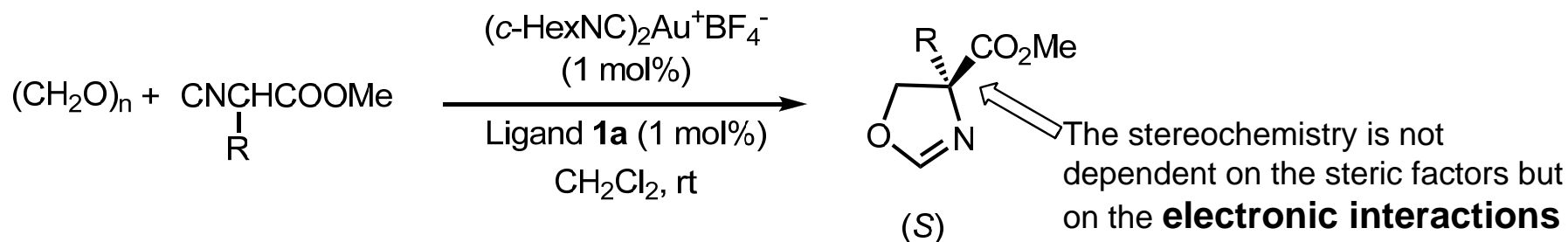
Aldehyde	Yield (%)	<i>trans</i> : <i>cis</i>	ee (<i>trans</i> , %)	ee (<i>cis</i> , %)
PhCHO	98	89:11	96	49
MeCHO	100	84:16	72	44
<i>i</i> -PrCHO	99	98:2	92	—
<i>c</i> -HexCHO	95	97:3	90	—
<i>t</i> -BuCHO	100	100:0	97	—
(<i>E</i>)- <i>n</i> -PrCH=CHCHO	83	81:19	84	52

Aldol Reaction of Aldehydes with α -isocyanocarboxylates

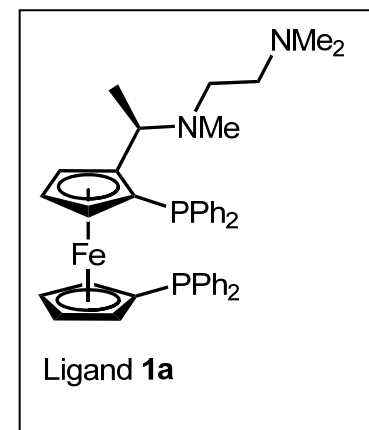


R	Yield (%)	<i>trans:cis</i>	<i>ee (trans, %)</i>	<i>ee (cis, %)</i>
H	91	90:10	91	4
Me	95	82:18	92	44
<i>i</i> -Pr	95	50:50	88	48

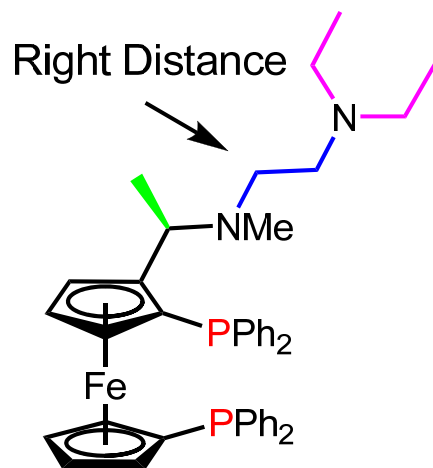
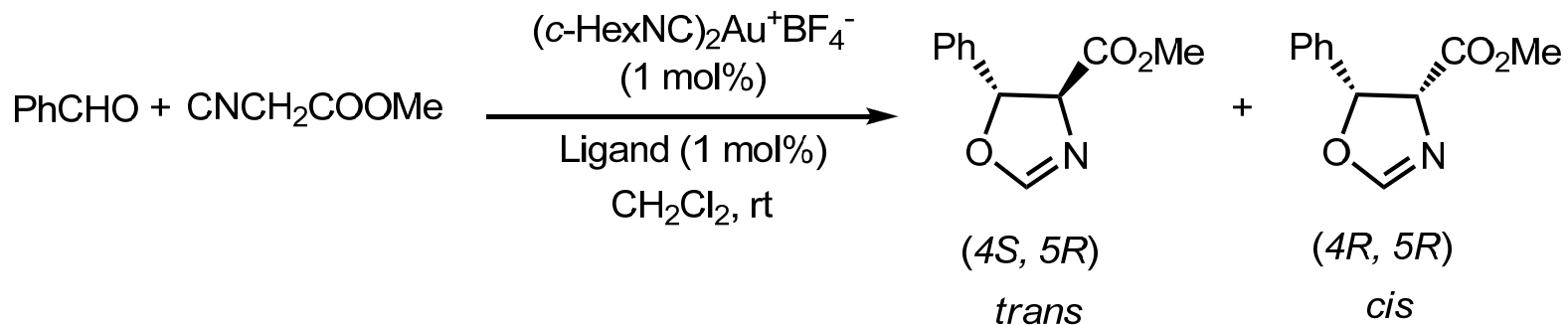
Aldol Reaction of Different Isocyanocarboxylates



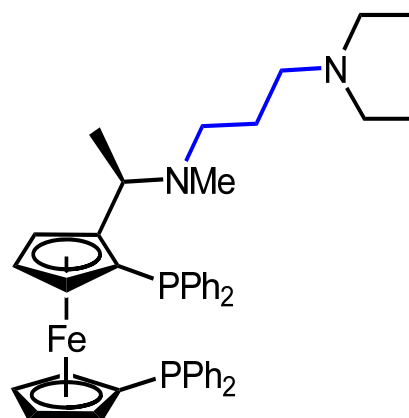
R	Yield (%)	ee (%)
H	99	52
Me	100	64
Et	89	70
<i>i</i> -Pr	99	71
Ph	75	67



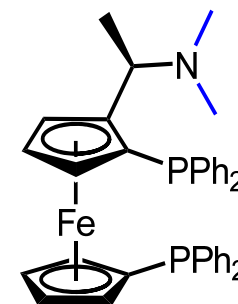
Aldol Reaction with Different Ligands



Yield: 98%
trans:cis: 89:11
ee (trans): 96%
ee (cis): 49%

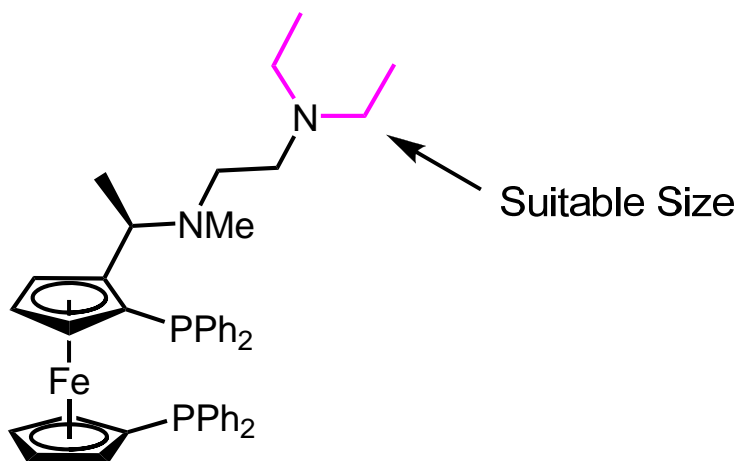
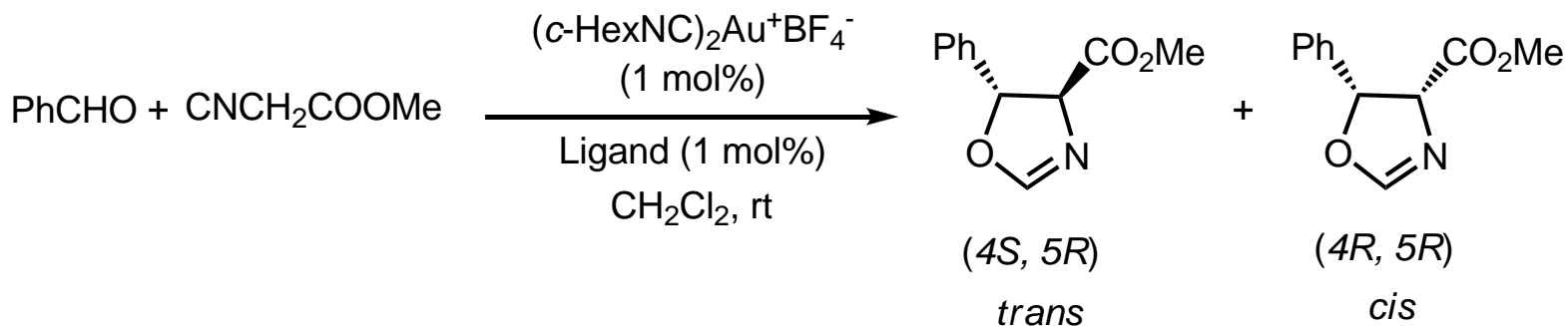


ee (trans): 26%

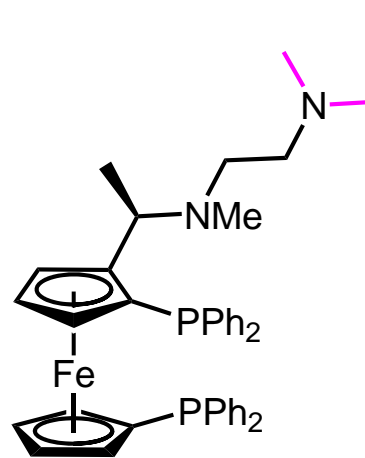


ee (trans): ~0%

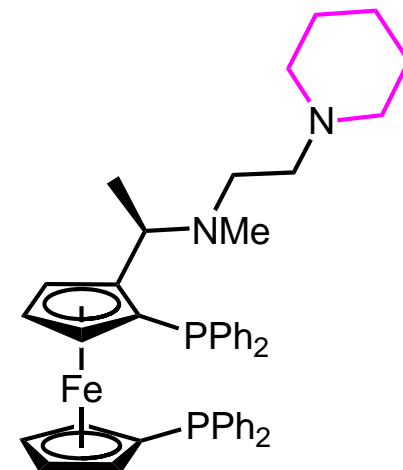
Aldol Reaction with Different Ligands



Yield: 98%
trans:cis: 89:11
ee (trans): 96%
ee (cis): 49%

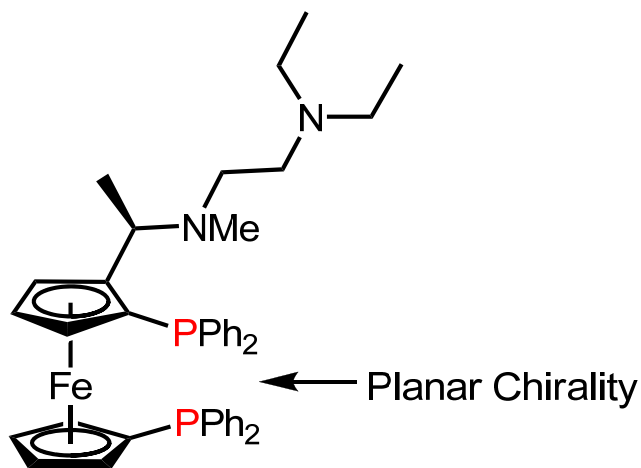
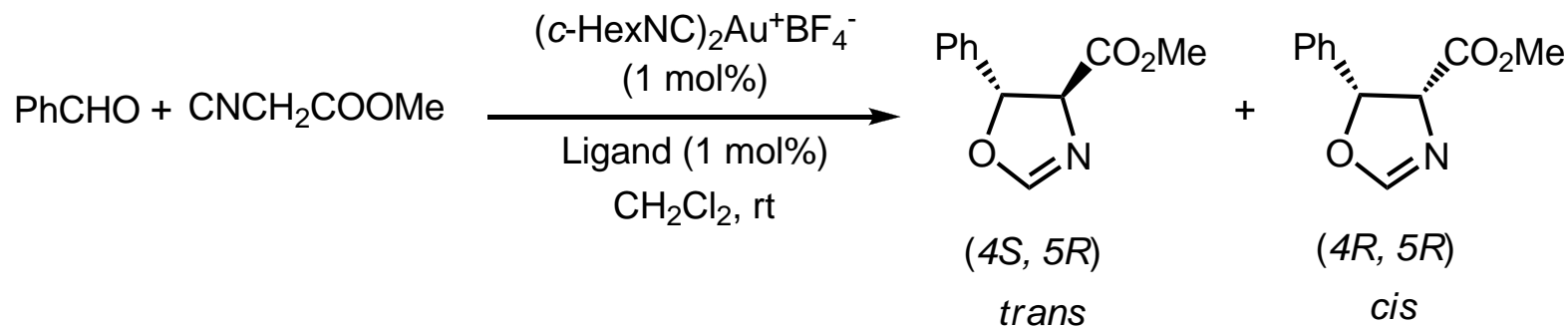


Yield: 91%
trans:cis: 90:10
ee (trans): 94%
ee (cis): - 4%



Yield: 94%
trans:cis: 94:6
ee (trans): 95%
ee (cis): 49%

Aldol Reaction with Different Ligands

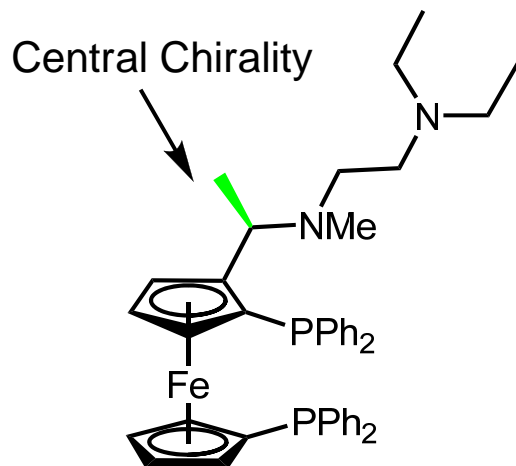
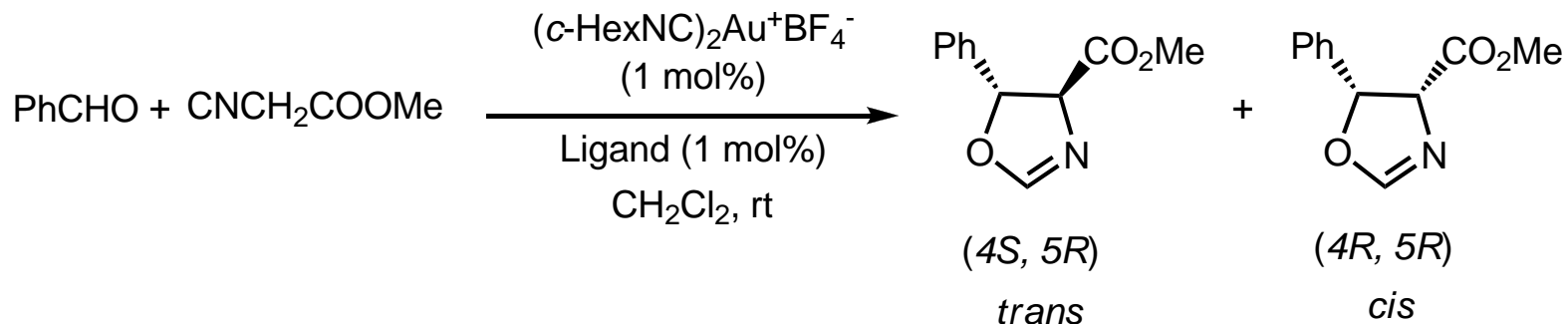


Yield: 98%
trans:cis: 89:11
ee (trans): 96%
ee (cis): 49%

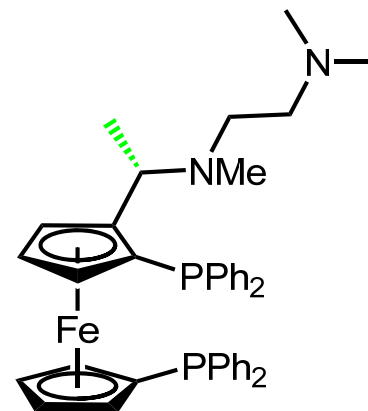
Yield: 74%
trans:cis: 72:28
ee (trans): 0%
ee (cis): 0%

Yield: 63%
trans:cis: 69:31
ee (trans): 0%
ee (cis): 0%

Aldol Reaction with Different Ligands

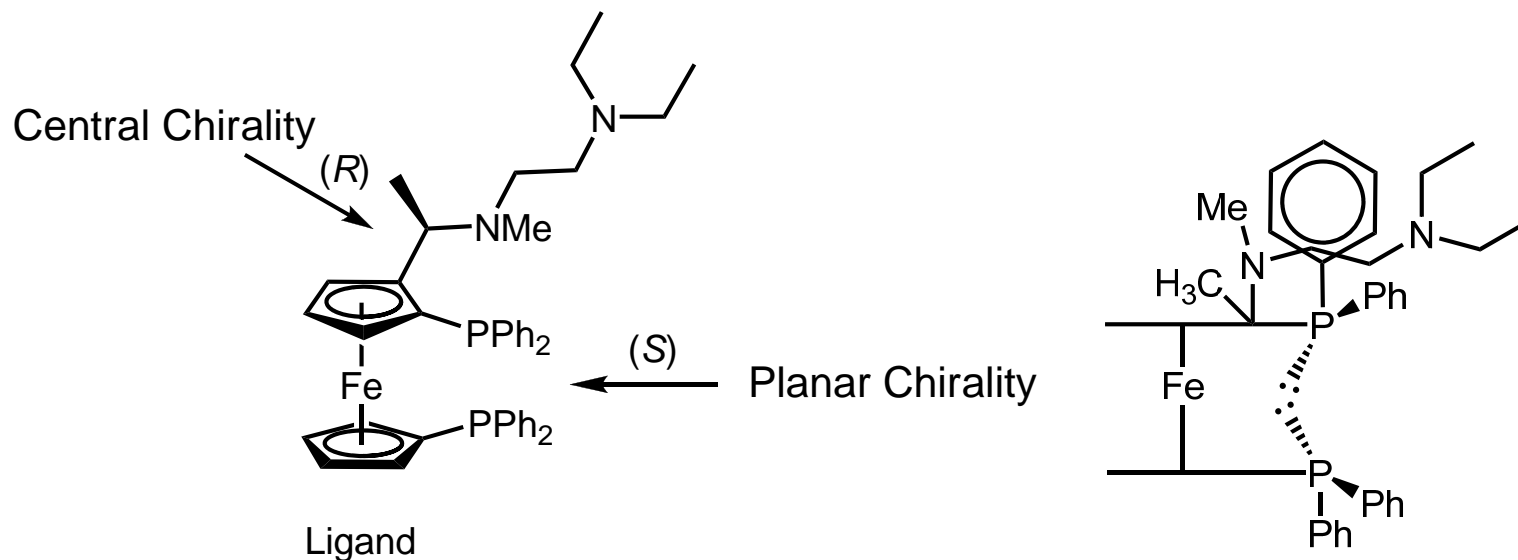
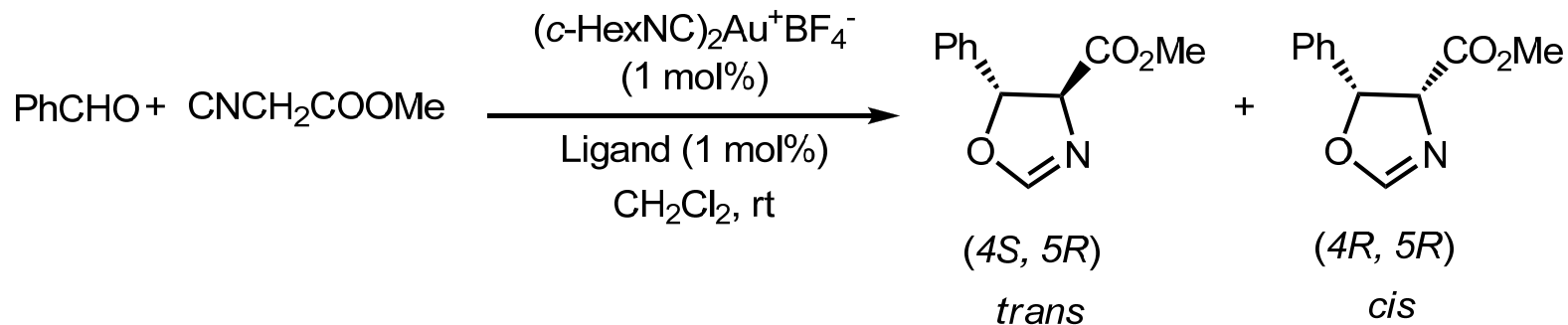


Yield: 98%
trans:cis: 89:11
ee (trans): 96%
ee (cis): 49%

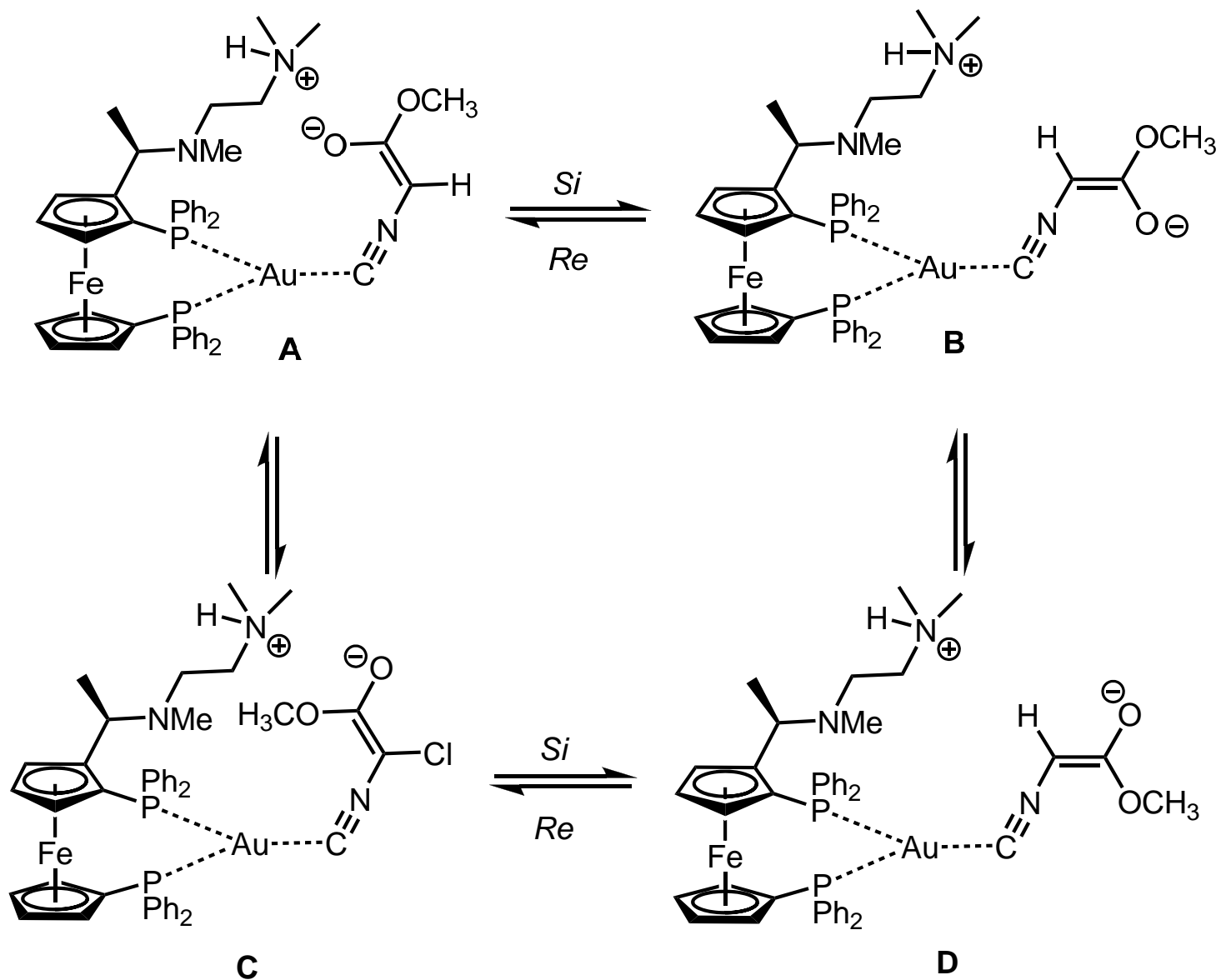


Yield: 90%
trans:cis: 84:16
ee (trans): 41%
ee (cis): -20%

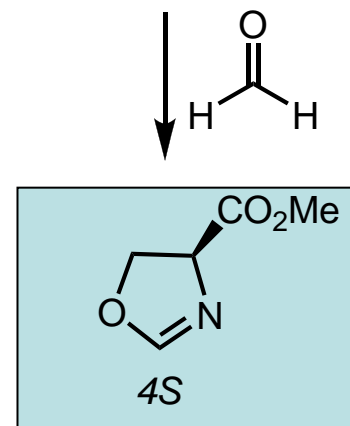
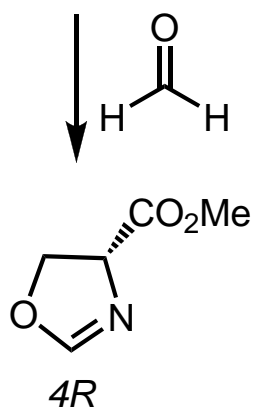
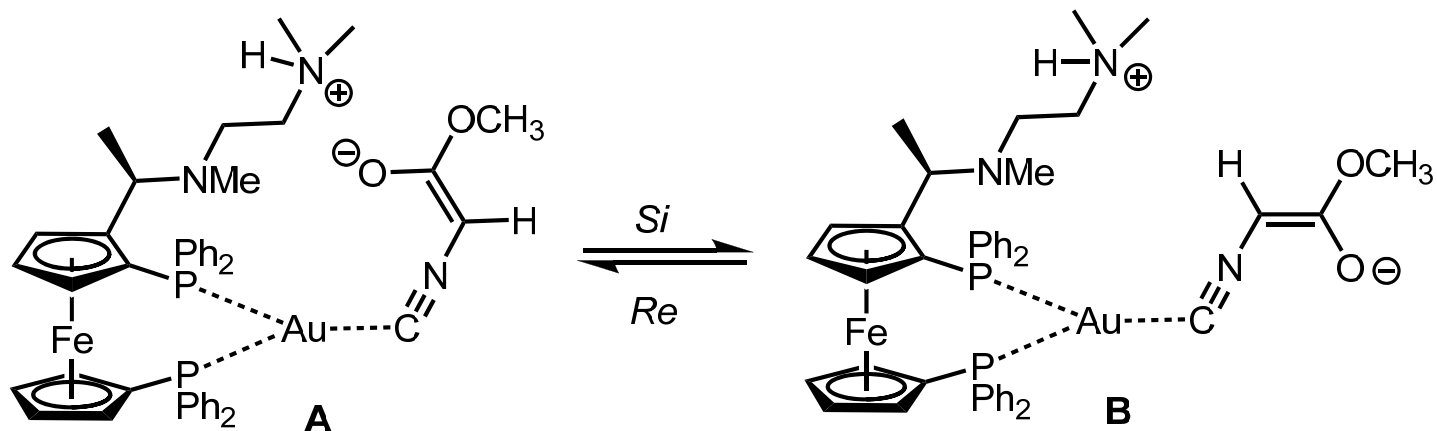
Internal Chiral Cooperativity in the Ligands



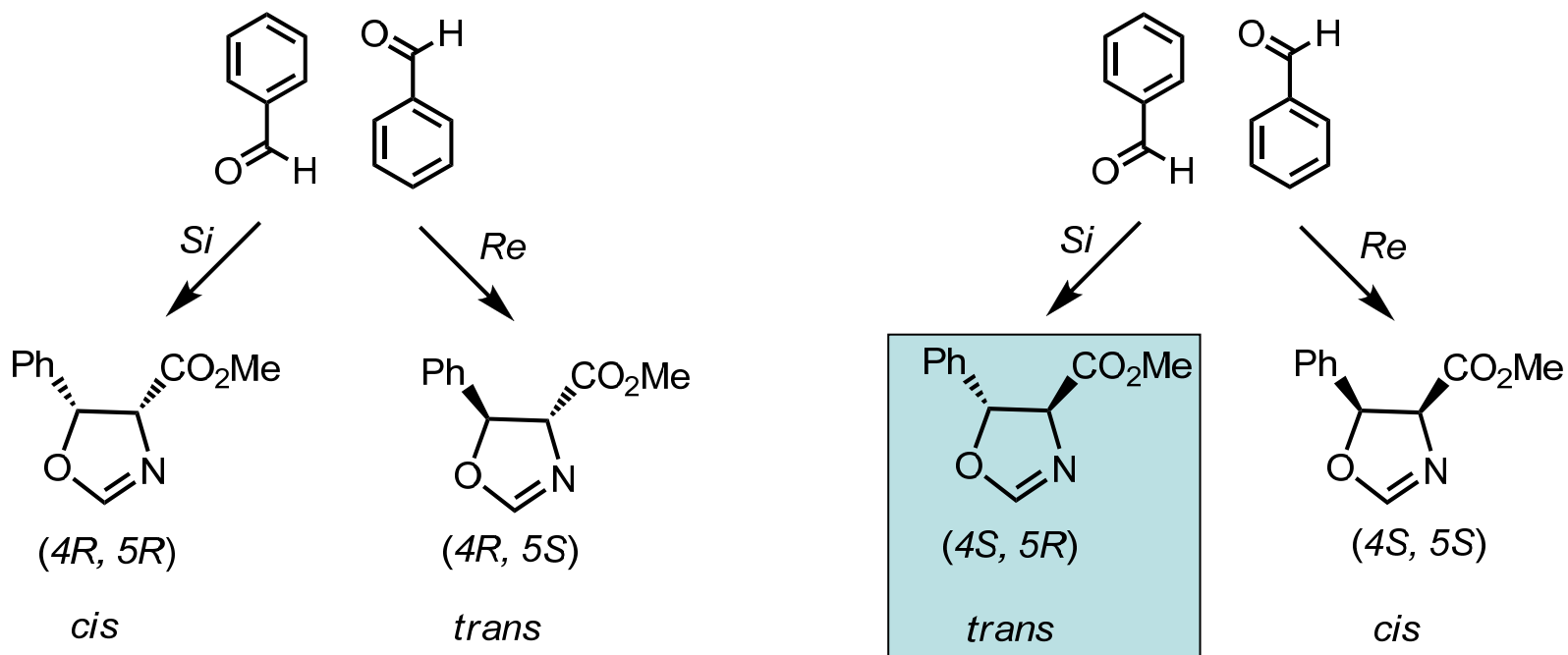
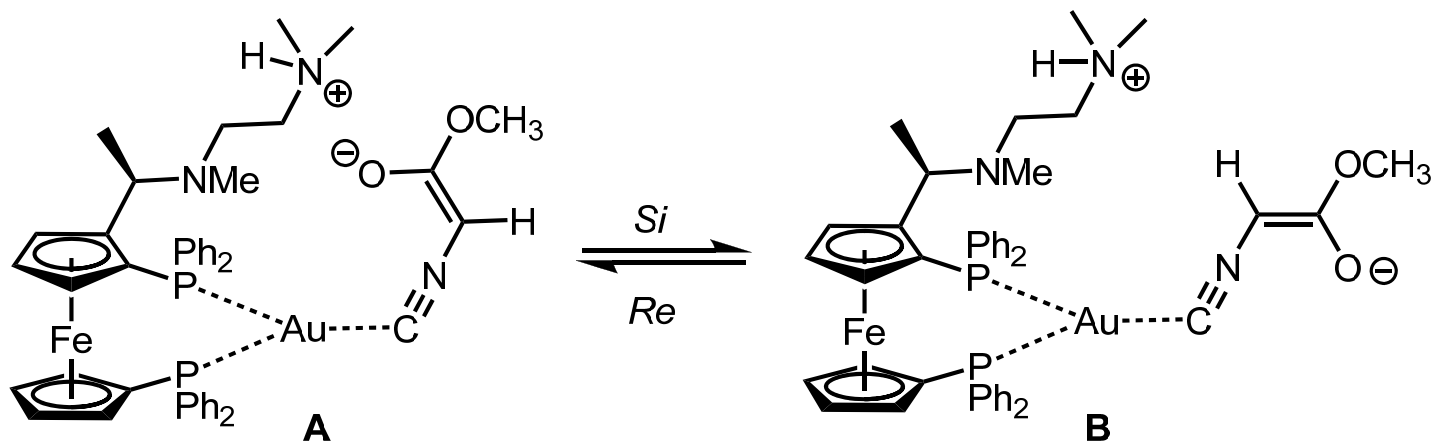
Transition State



Transition State



Transition State for Aldol Reaction



Enantioselective Reactions

Hydrofunctionalization

Hydrogenation

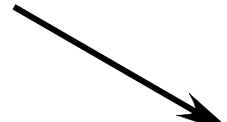
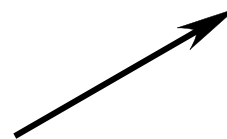
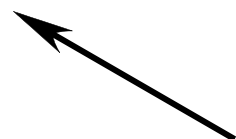


Chiral Au(I) Complex

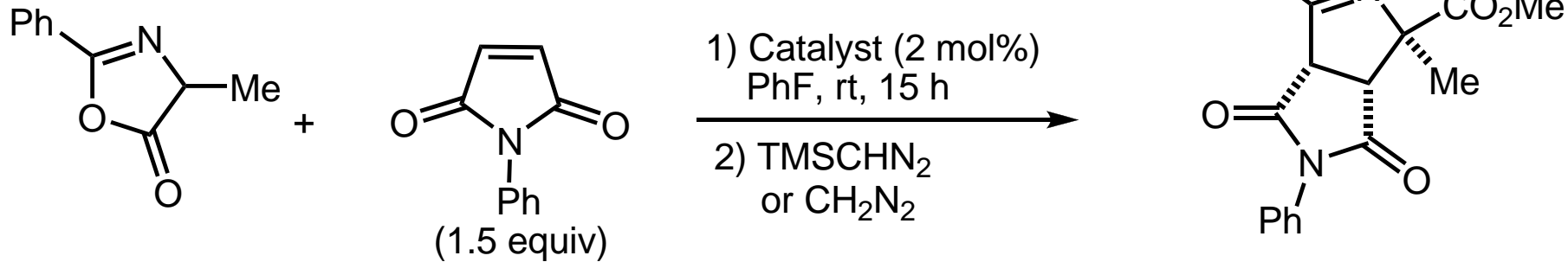
Cyclopropanation

Aldol Reaction

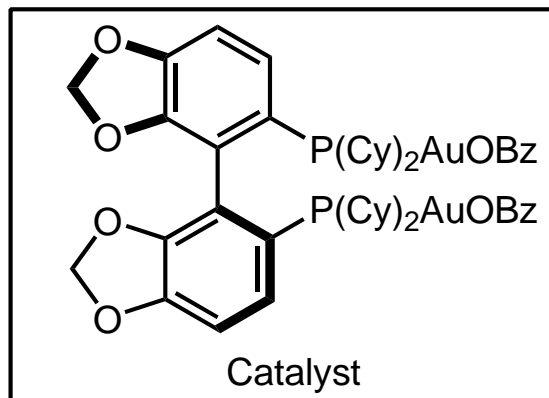
1,3-Dipolar Cycloaddition



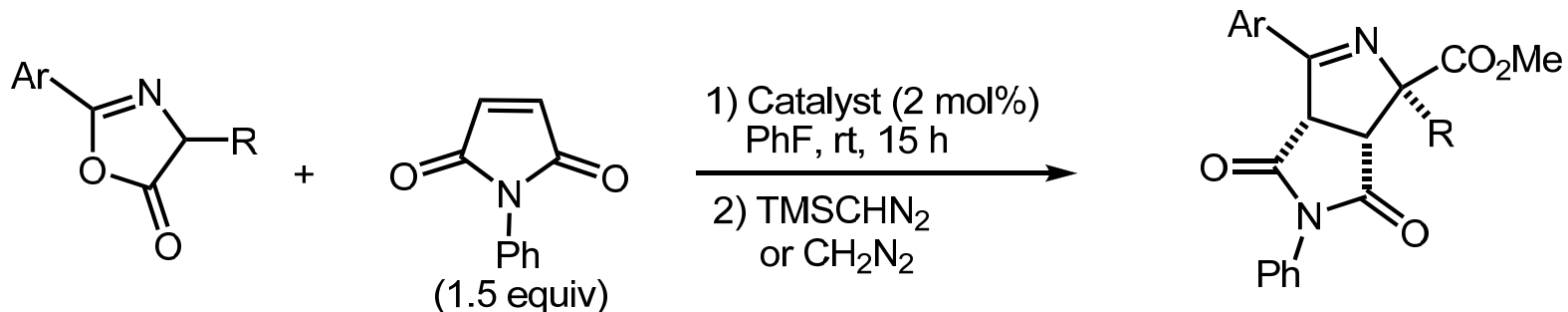
1,3-Dipolar Cycloaddition



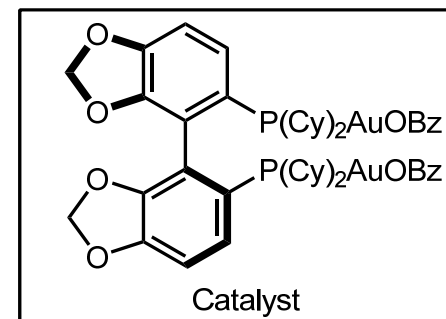
Yield: 76%
ee: 95%



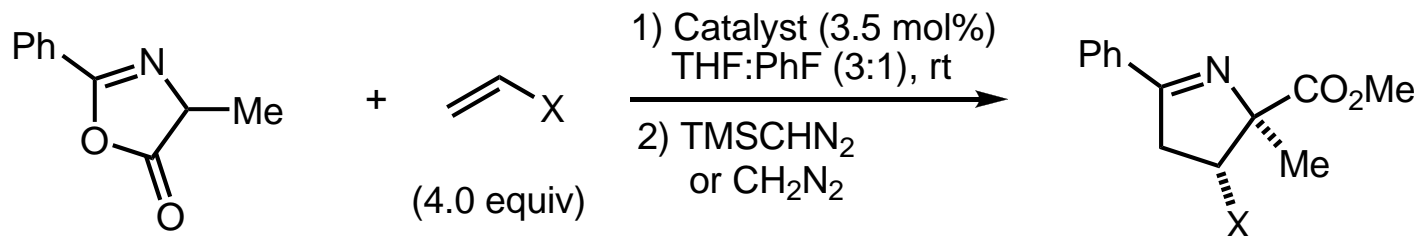
Substrate Scope for Cycloaddition



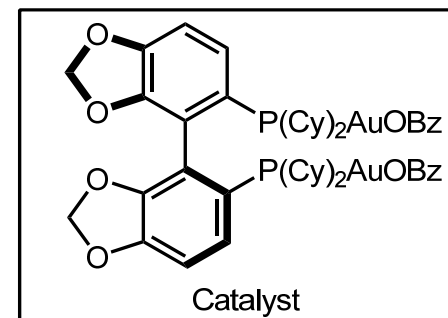
R	Ar	Yield (%)	ee (%)
Me	<i>p</i> -MeO-C ₆ H ₄	77	95
Me	<i>p</i> -Br-C ₆ H ₄	75	93
Me	<i>p</i> -Cl-C ₆ H ₄	72	92
Me	<i>p</i> -NO ₂ -C ₆ H ₄	98	91
Me	<i>o</i> -Me-C ₆ H ₄	73	86
H	Ph	84	81
Allyl	Ph	86	87
Bn	Ph	71	68
Ph	Ph	35	78



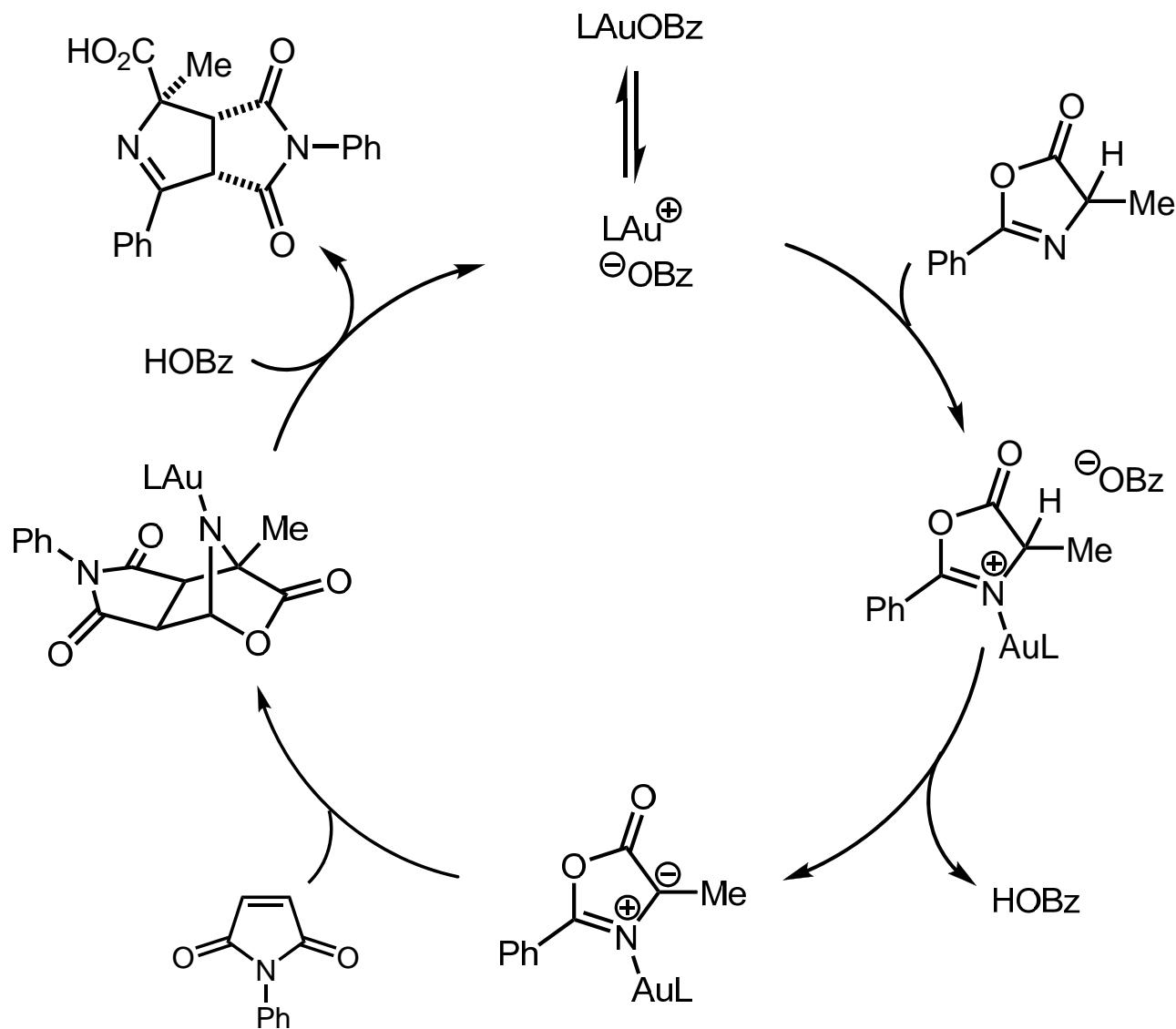
Cycloaddition with Acyclic Alkenes



X	time (h)	Yield (%)	ee (%)
CO ₂ <i>t</i> -Bu	24	56	99
CO ₂ Et	14	66	90
CO ₂ Me	14	89	93



Mechanism for 1,3-Dipolar Cycloaddition



Enantioselective Reactions



Hydrofunctionalization

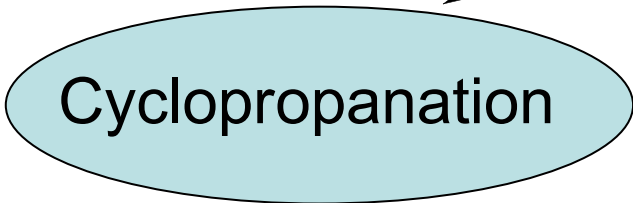
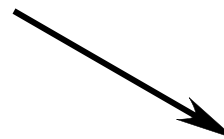
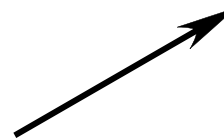
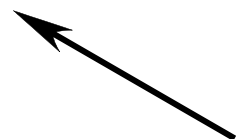
Hydrogenation

Chiral Au(I) Complex

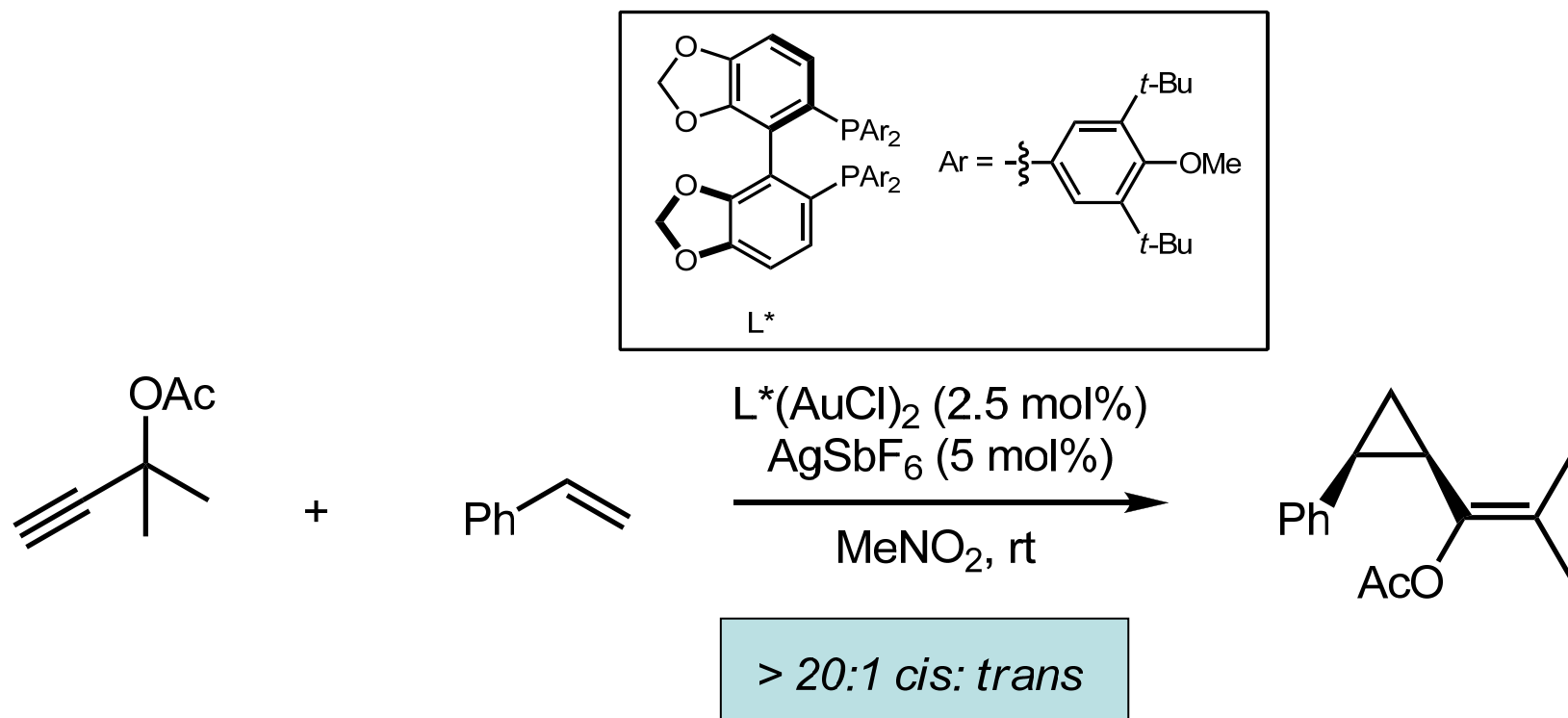
Cyclopropanation

Aldol Reaction

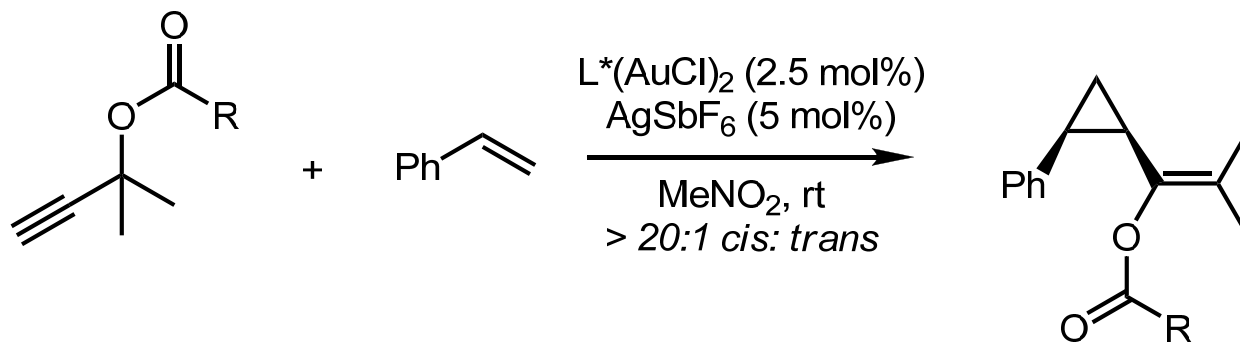
1,3-Dipolar Cycloaddition



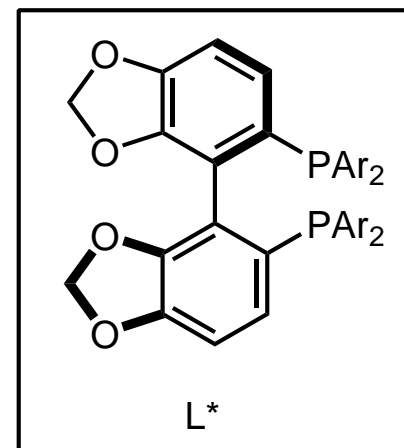
Olefin Cyclopropanation Reaction



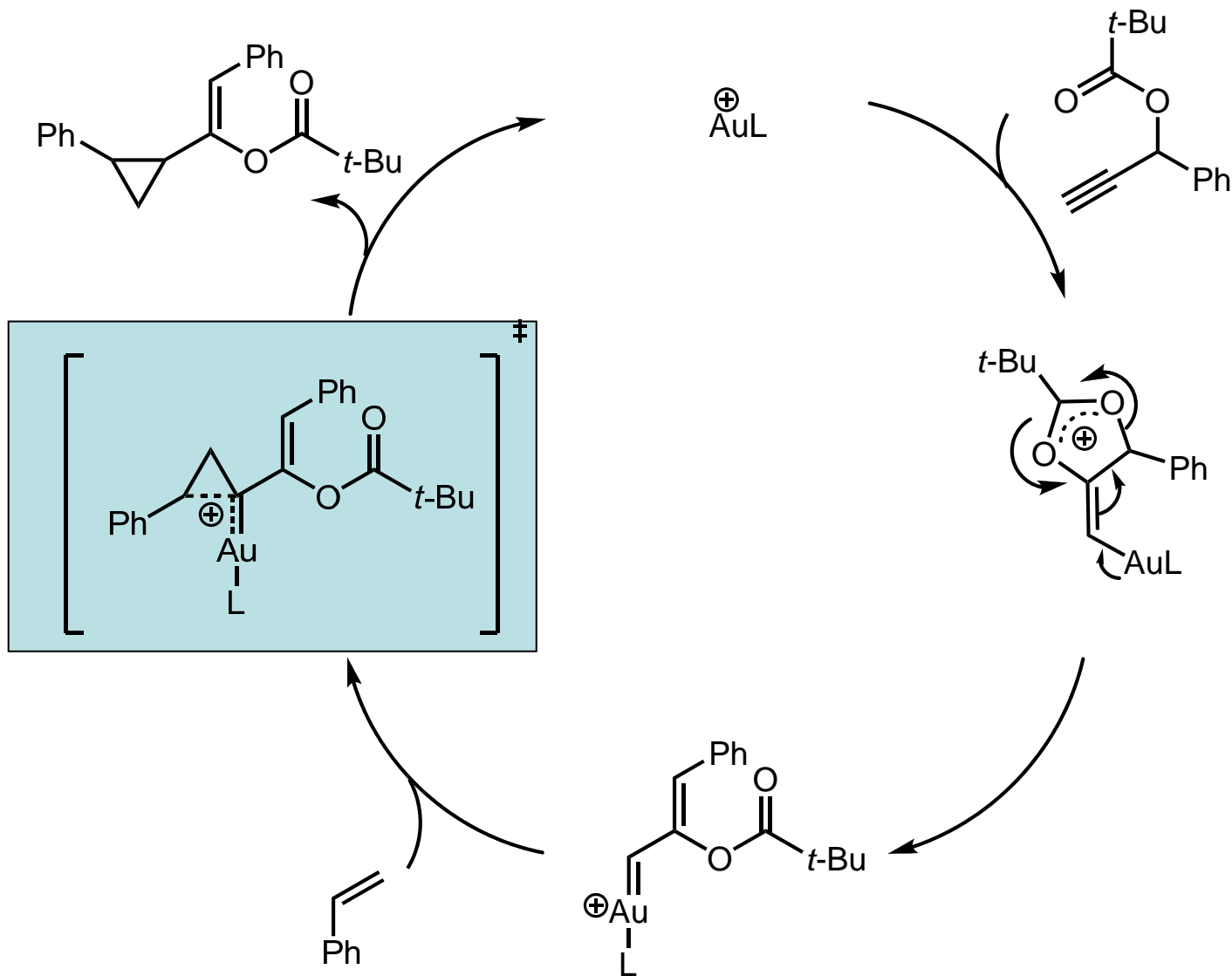
Olefin Cyclopropanation



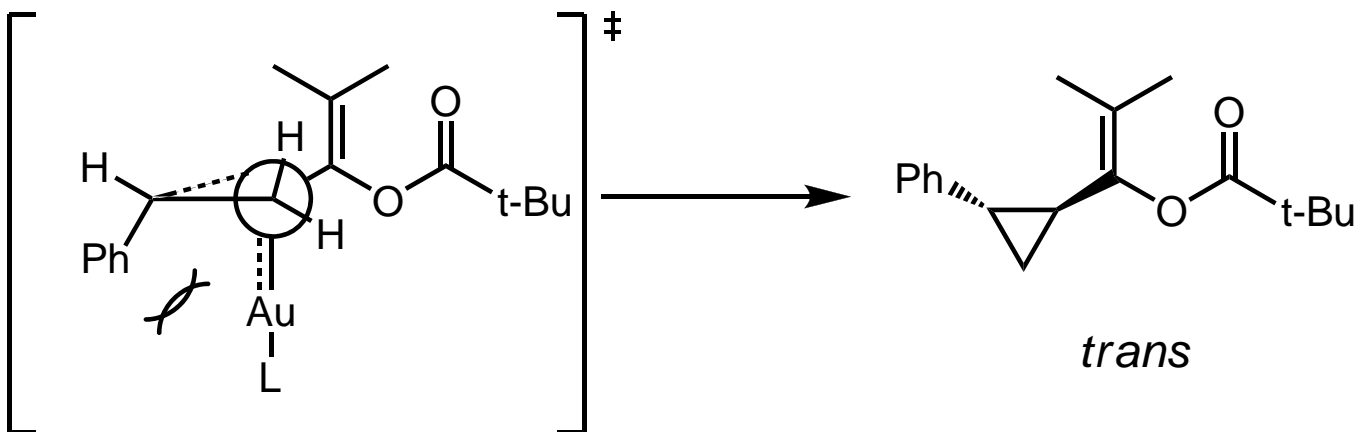
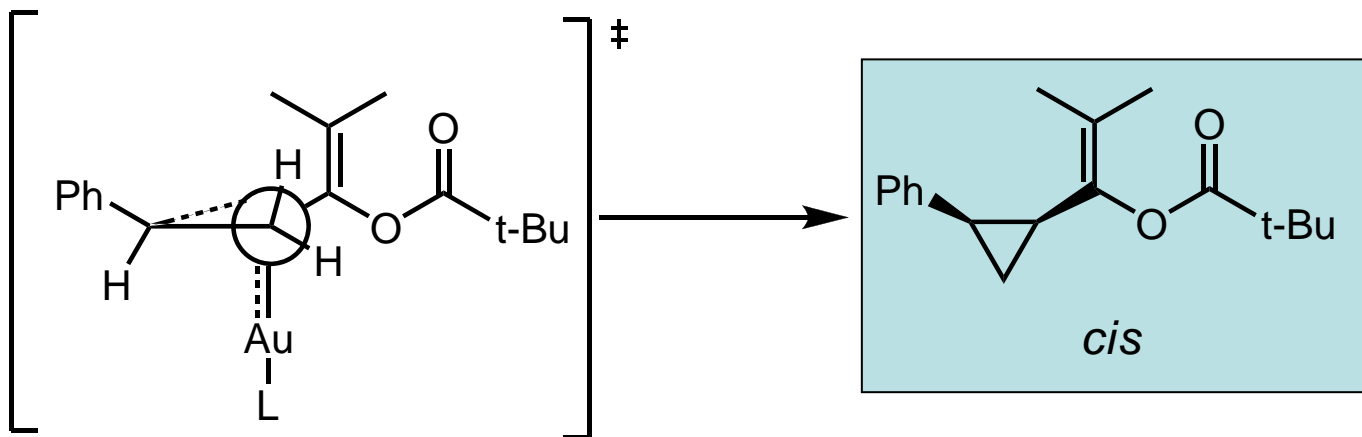
R	Yield (%)	ee (%)
Me	72	60
Ph	73	68
<i>t</i> -Bu	70	81



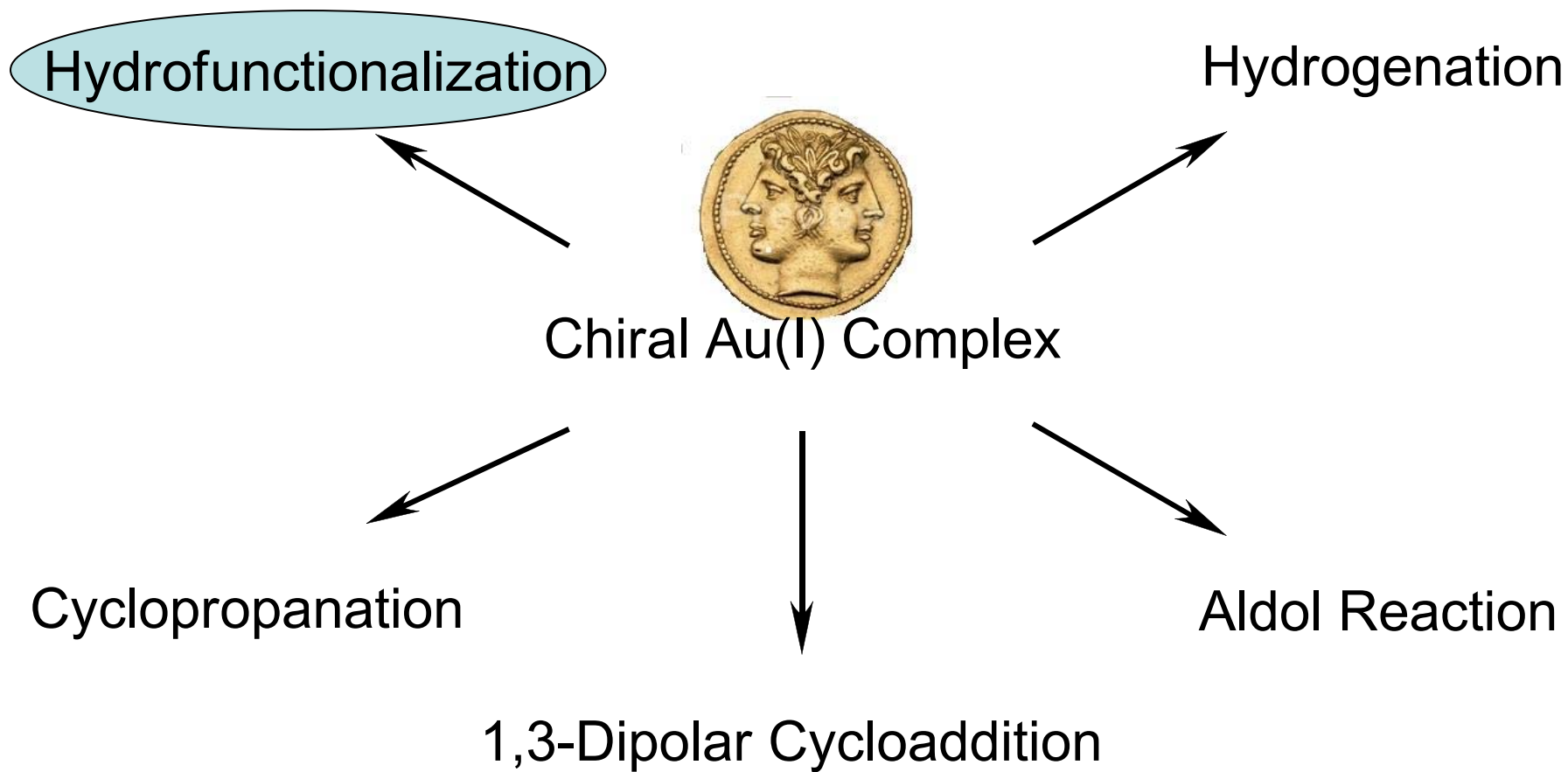
Mechanism for Cyclopropanation



Mechanism for Diastereoselectivity

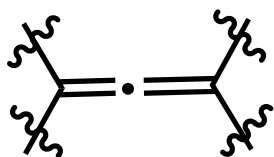
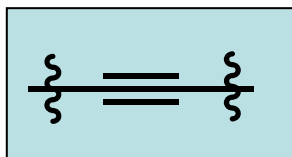


Enantioselective Reactions

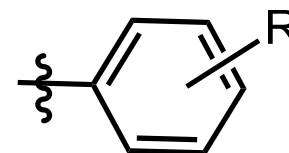
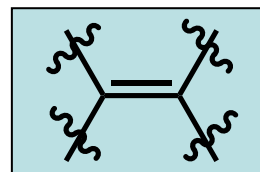


Classification for Intramolecular Hydrofunctionalization

Functional Groups 1



Functional Groups 2

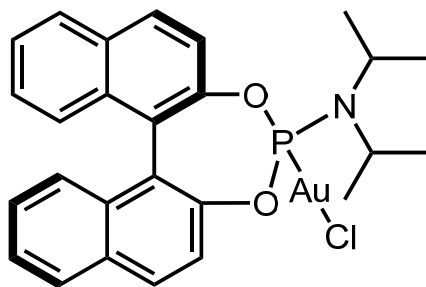
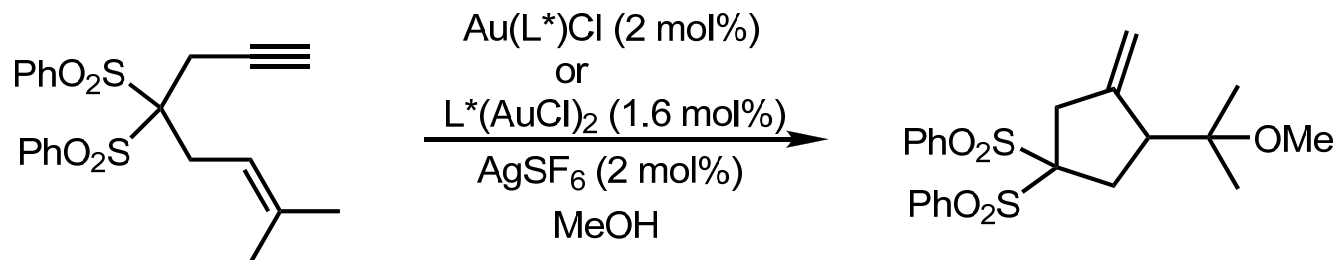


—OH

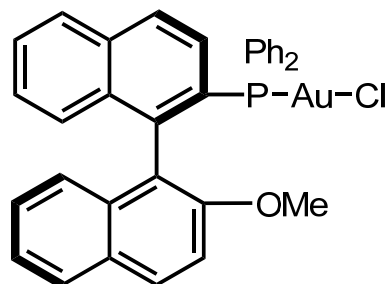
—COOH

—NHR

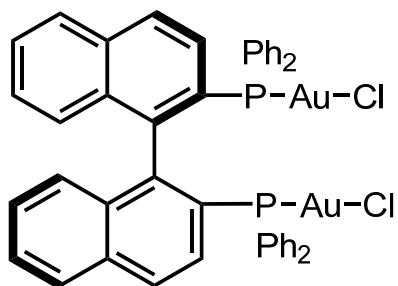
Catalysts for Cyclization of Enynes



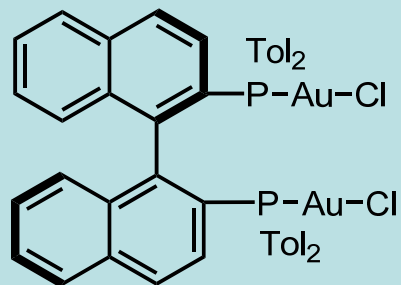
Yield: 60%
ee: <2%



Yield: 94%
ee: <2%

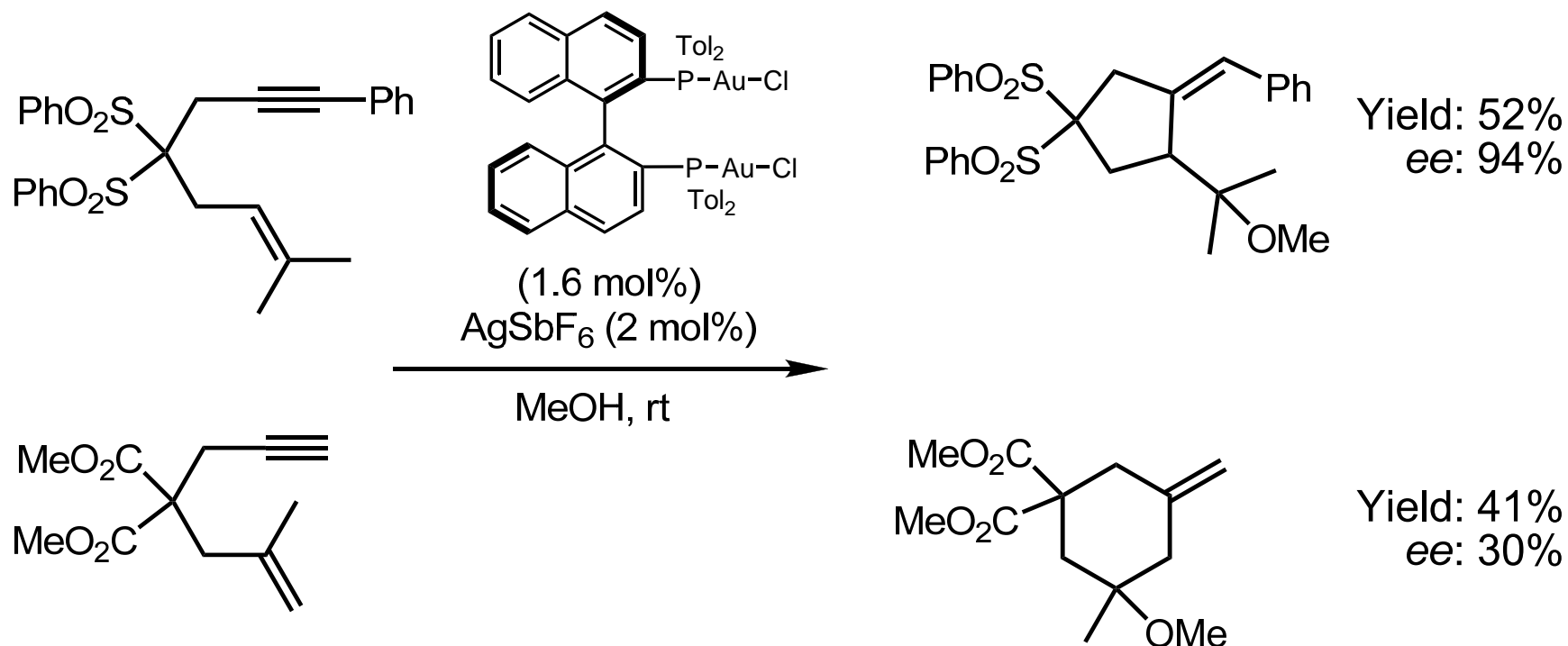


Yield: 98%
ee: -39%

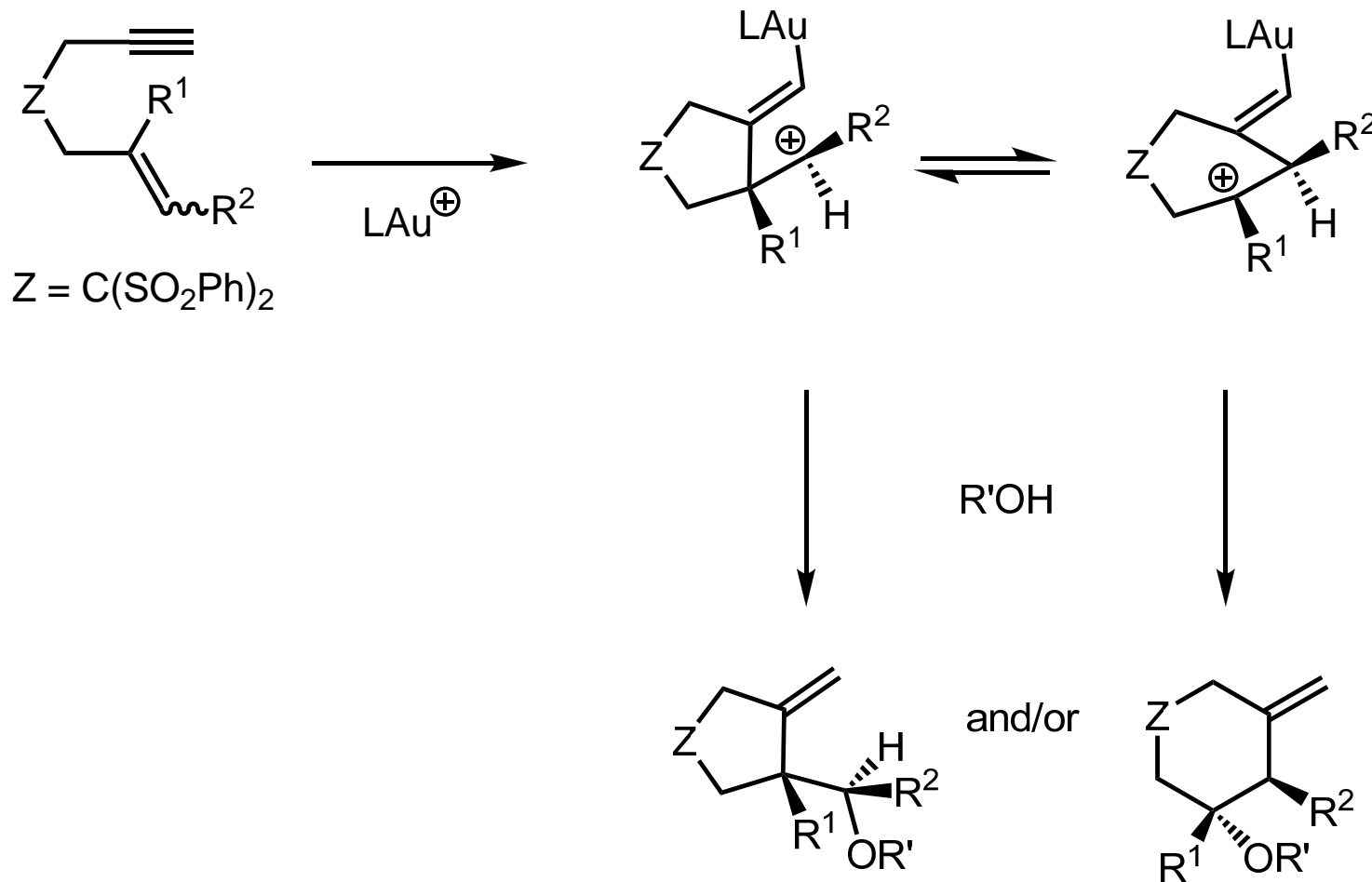


Yield: 93%
ee: -43%

Regioselectivity for Cyclization

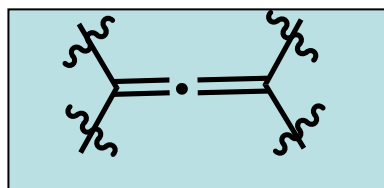


Mechanism for the Formation of Products

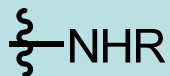
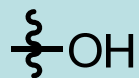
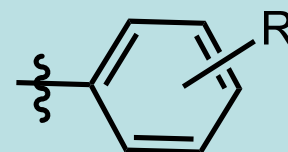
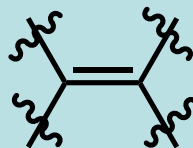


Classification for Intramolecular Hydrofunctionalization

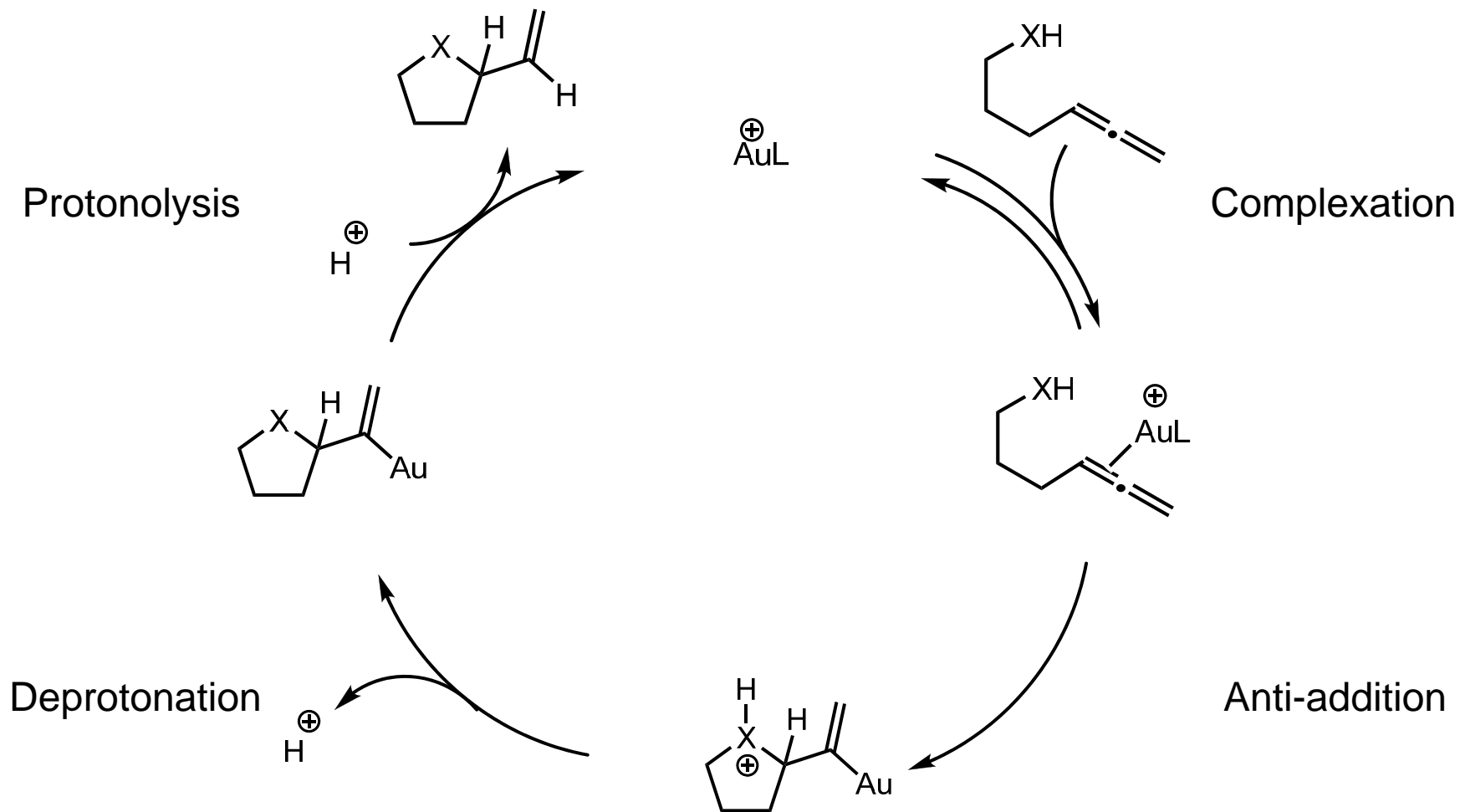
Functional Groups 1



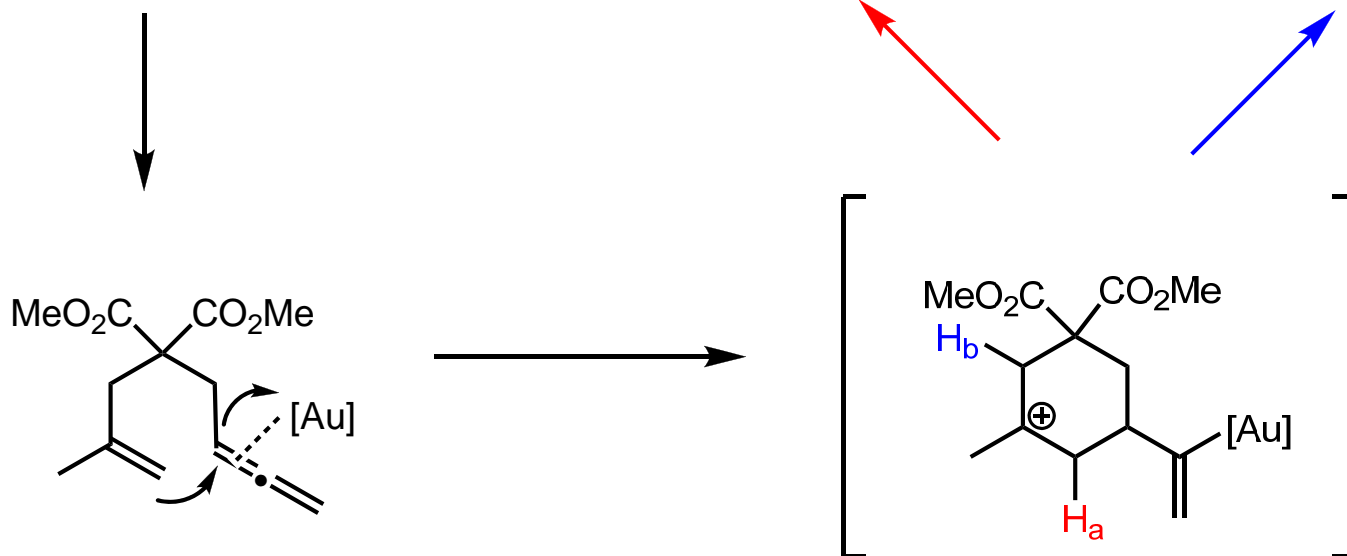
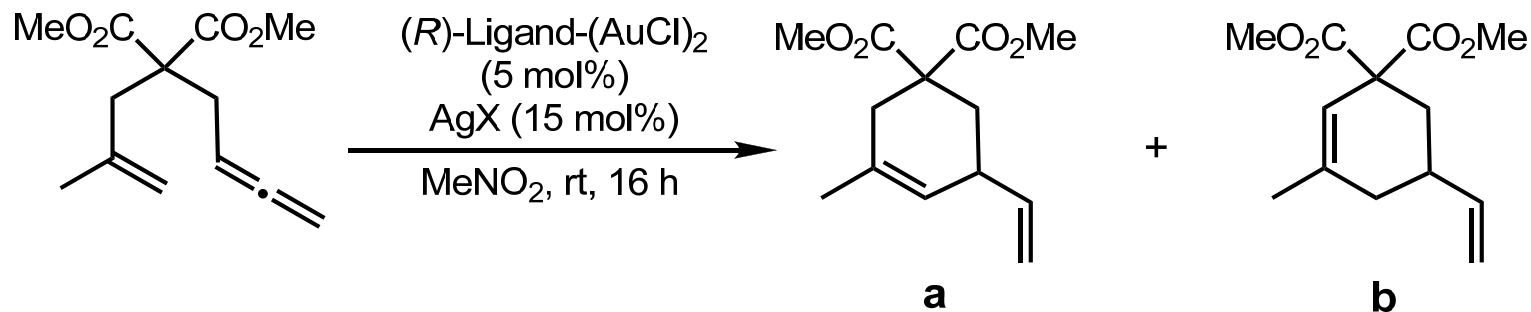
Functional Groups 2



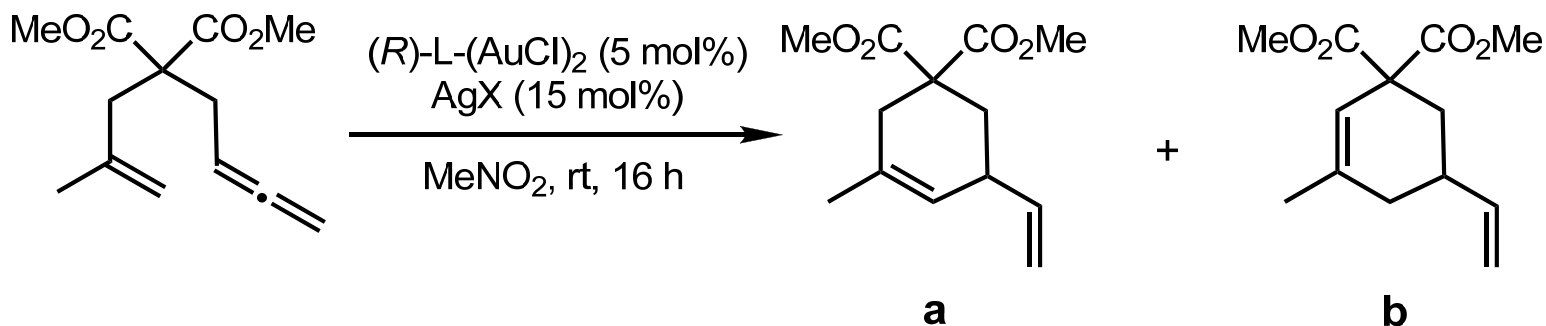
General Mechanism for Hydrofunctionalization of Allenes



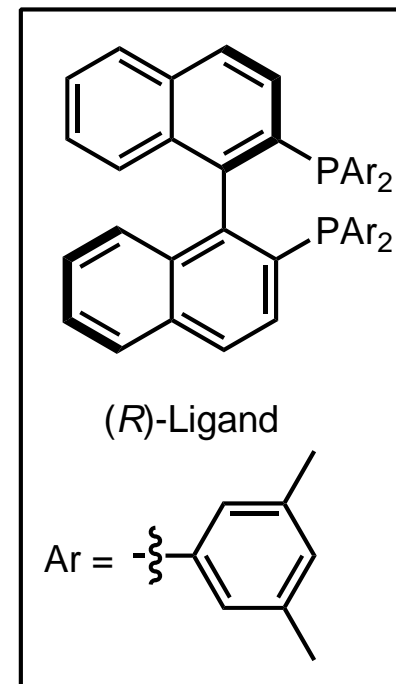
Hydroalkylation of Allenes



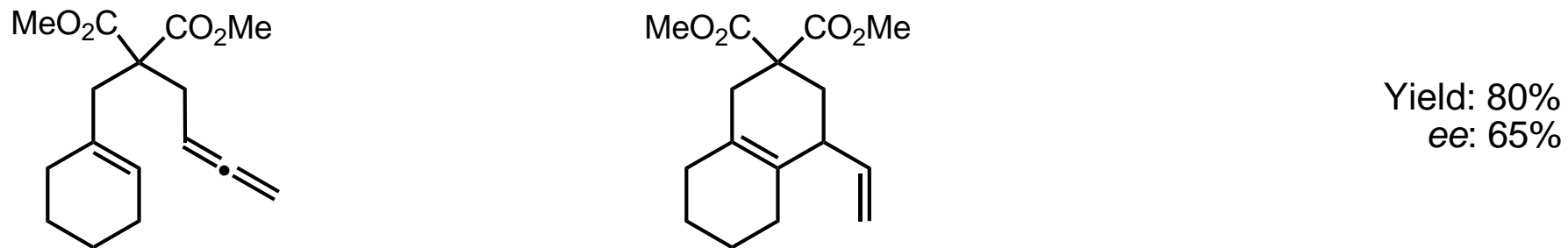
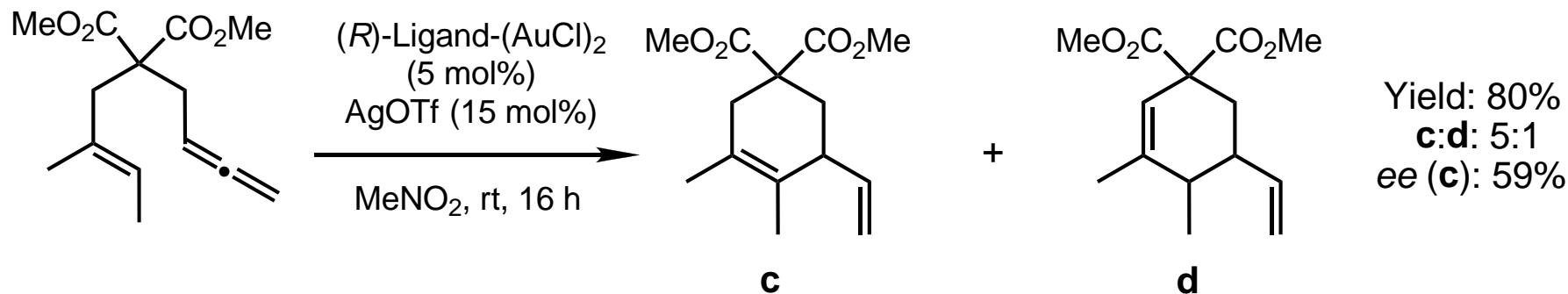
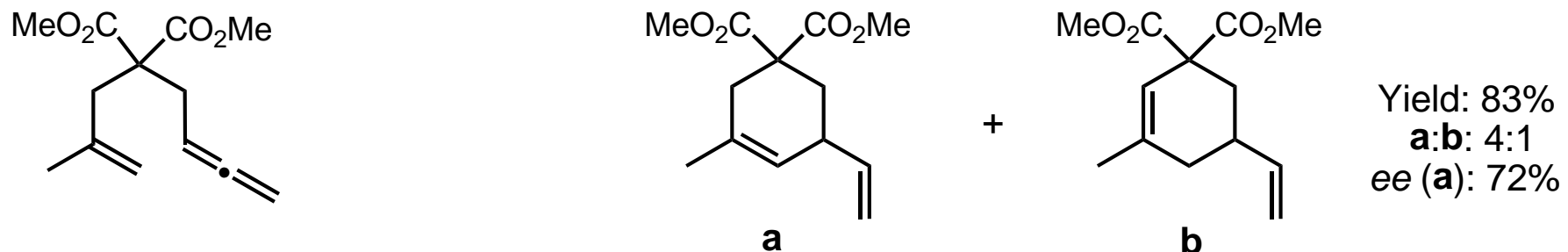
Counterion Effect in the Reaction



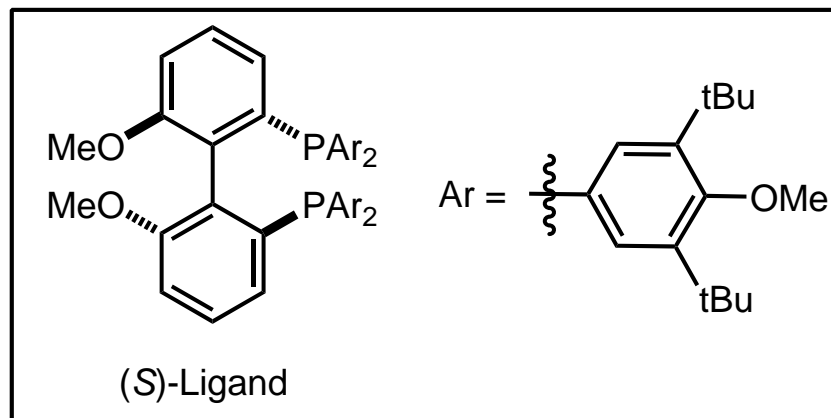
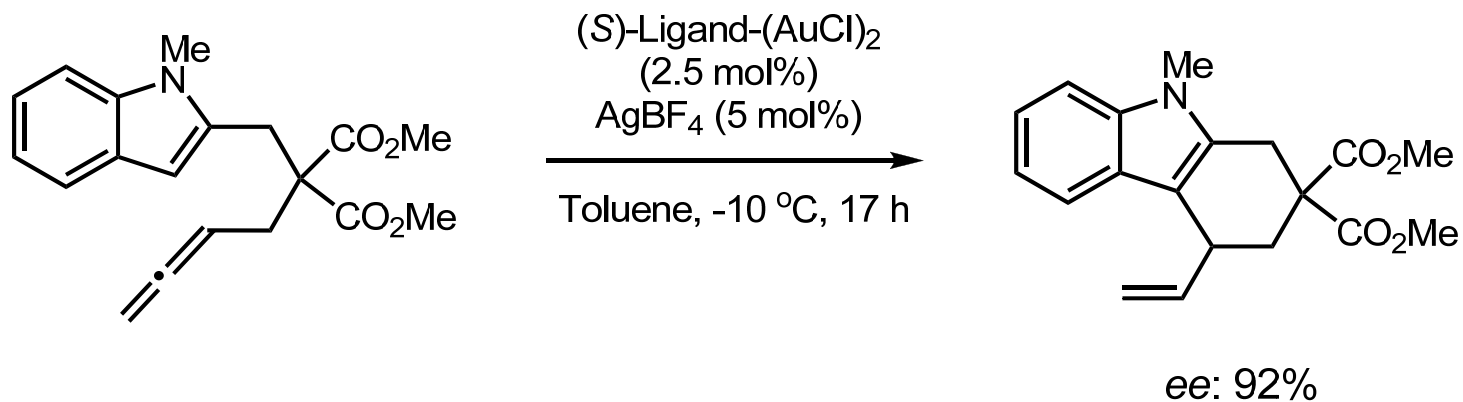
AgX	a:b	ee (a, %)
AgNTf ₂	4:1	65
AgSbF ₆	4:3	57
AgOTs	10:1	50
AgPF ₆	9:1	65
AgOTf	4:1	72



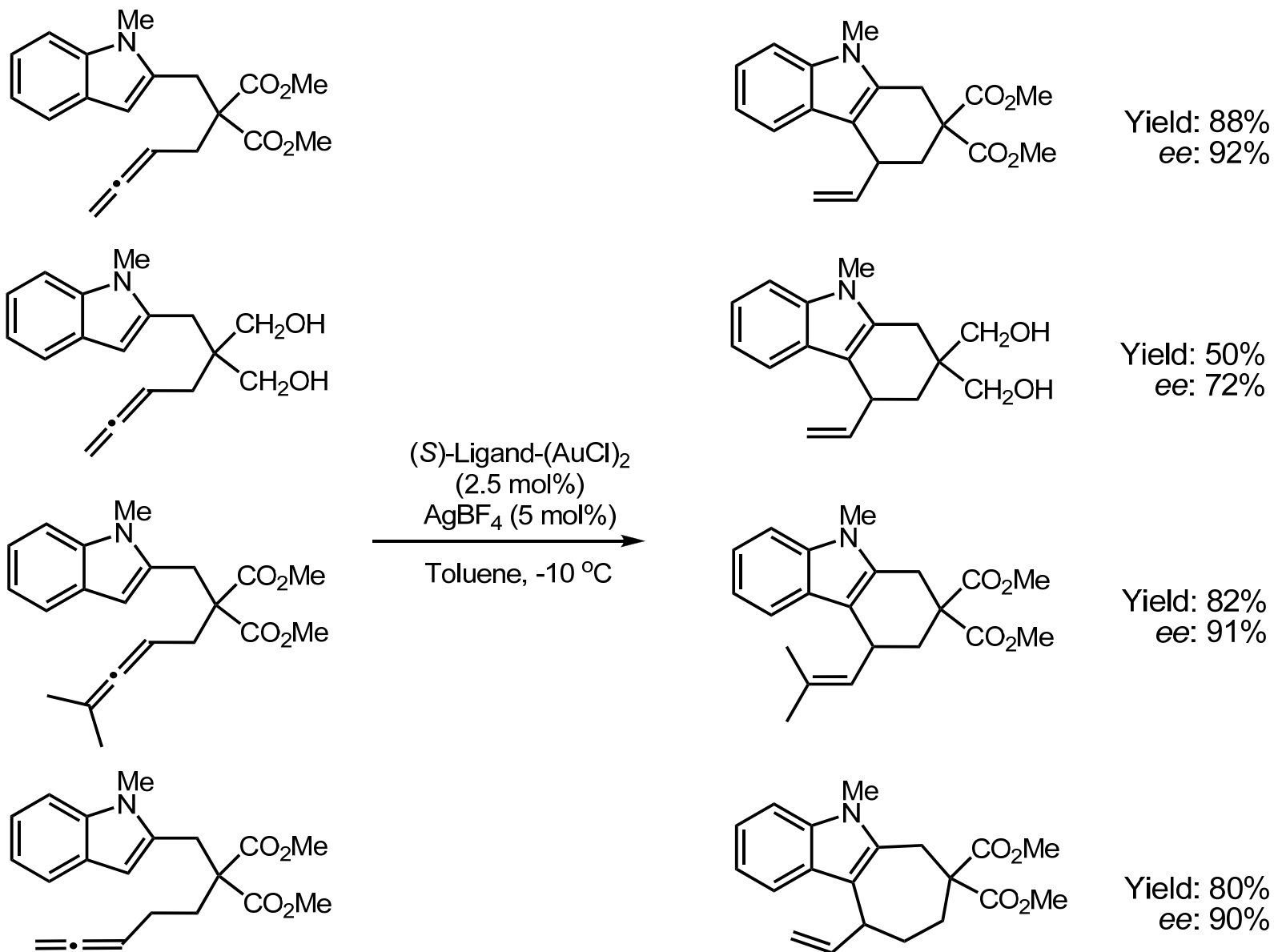
Substrate Scope for Hydroalkylation



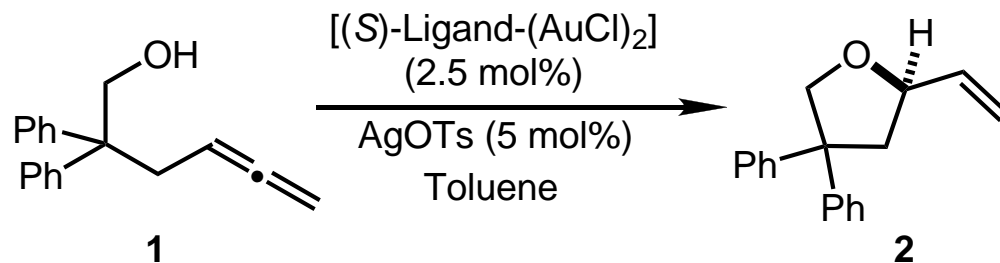
Hydroarylation of Allenes



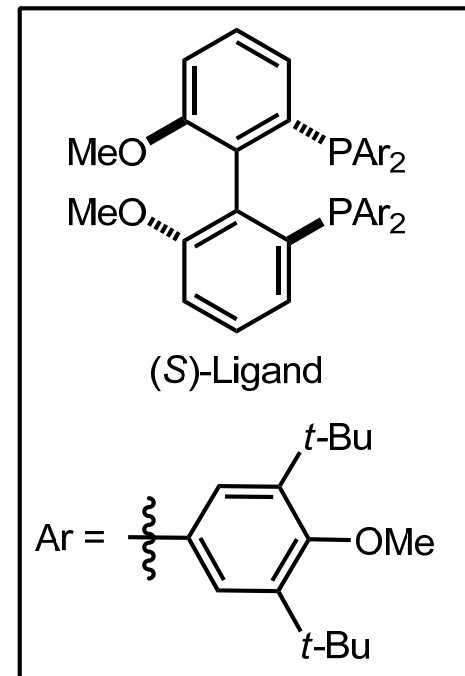
Substrate Scope for Hydroarylation



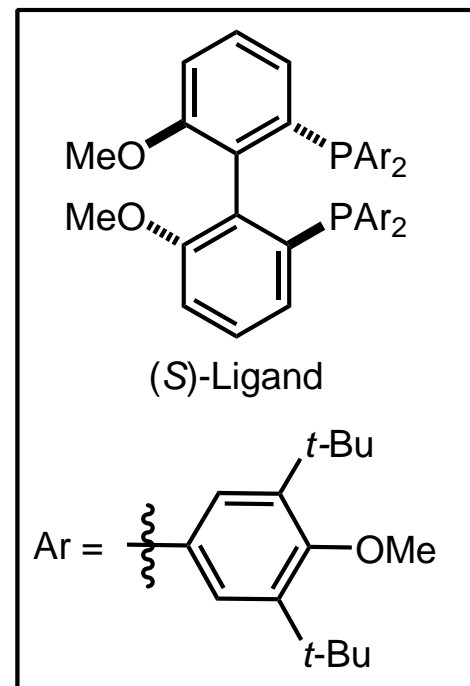
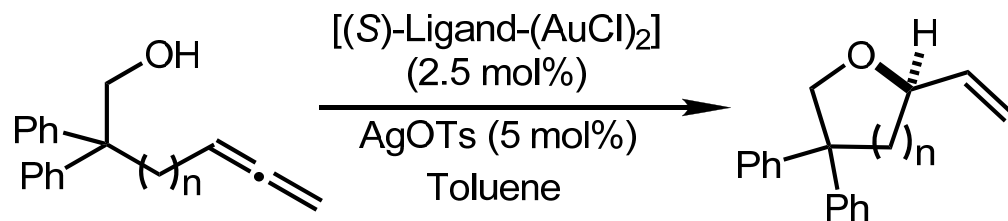
Intramolecular Hydroalkoxylation



Conc. 1 (mM)	T (°C)	t (h)	Yield (%)	ee (%)
125	25	< 0.1	73	86
13	25	4	73	90
13	-20	61	73	93
63	-20	18	76	93

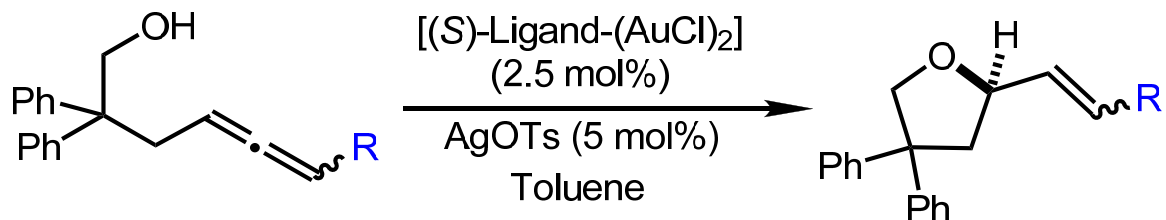


Intramolecular Hydroalkoxylation



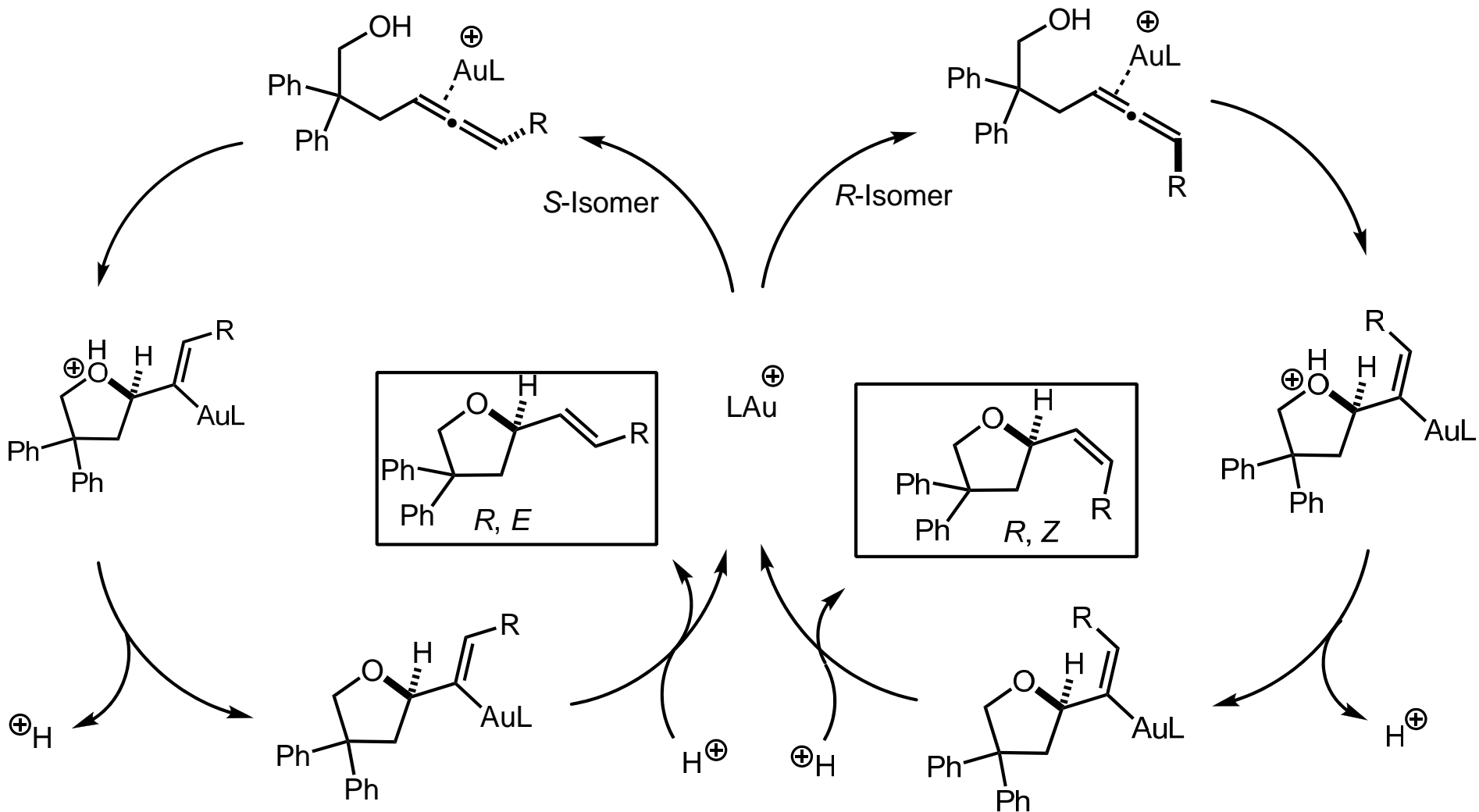
n	Yield (%)	ee (%)
1	67	93
2	96	88

Intramolecular Hydroalkoxylation

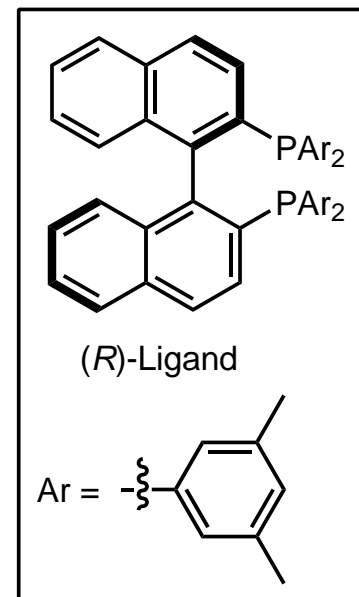
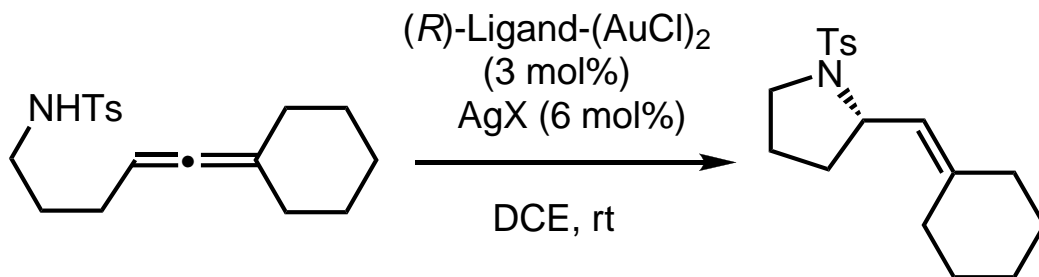


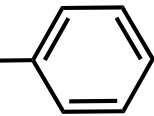
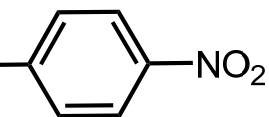
Substrate	Product	Z:E	Yield (%)	ee (%)
		1:1	96	97/99
		1:1	94	>95/>95
		>20:1	88	>95

Proposed Mechanism

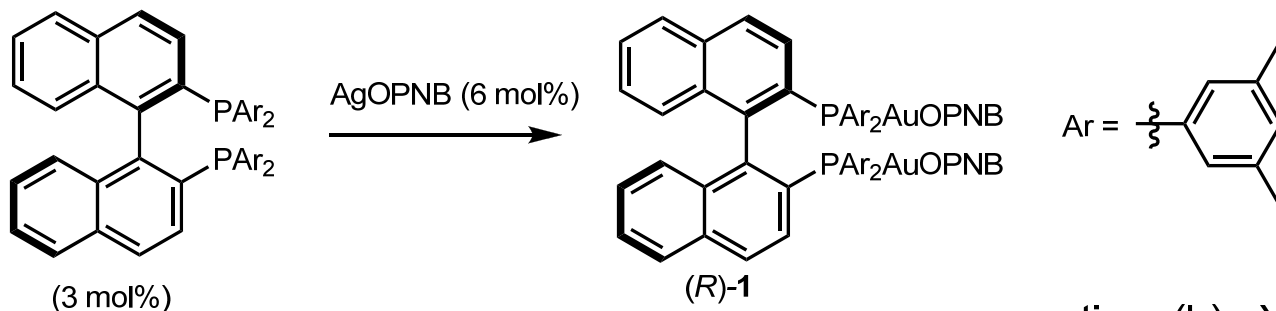
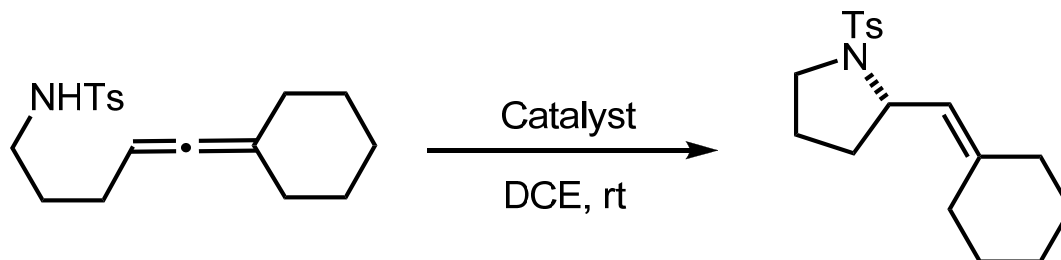


Counteranion Effect in Hydroamination



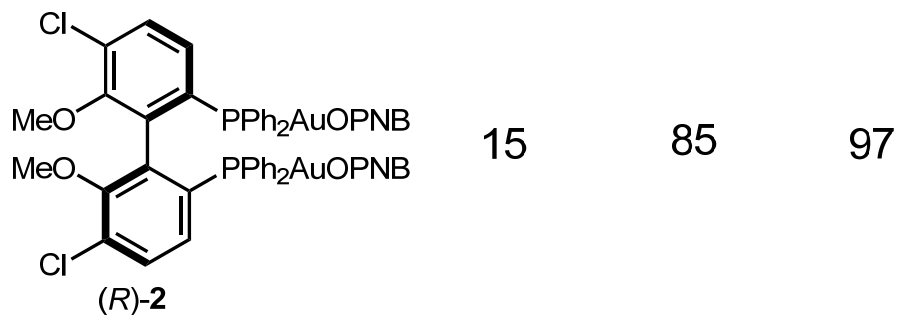
AgX	Time (h)	Yield (%)	ee (%)
AgBF_4	0.5	82	1
AgOOC - 	24	27	98
AgOOC - 	24	76	98

Intramolecular Hydroamination

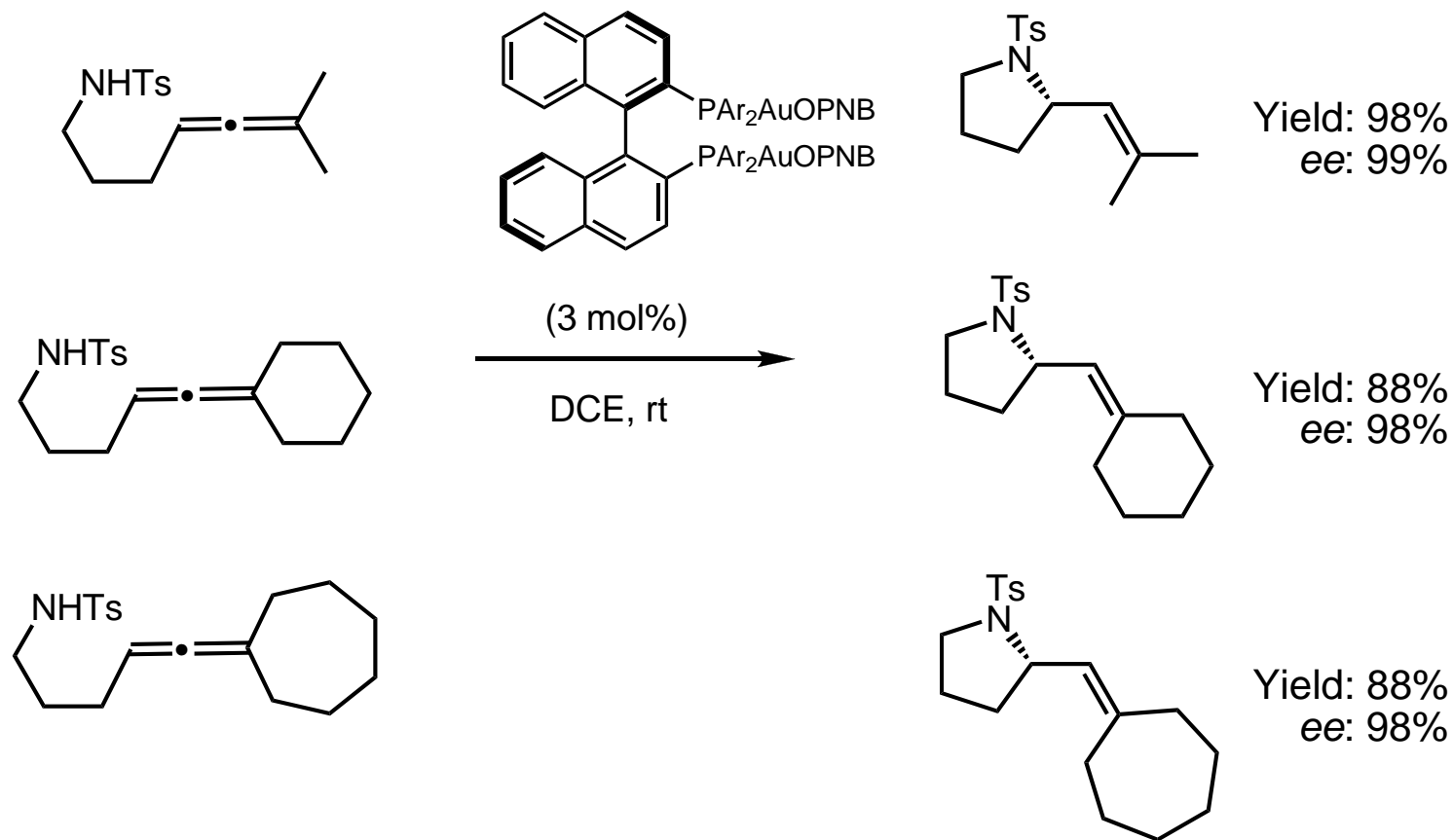


	time (h)	Yield (%)	ee (%)
<i>in situ</i> generated catalyst	24	76	98

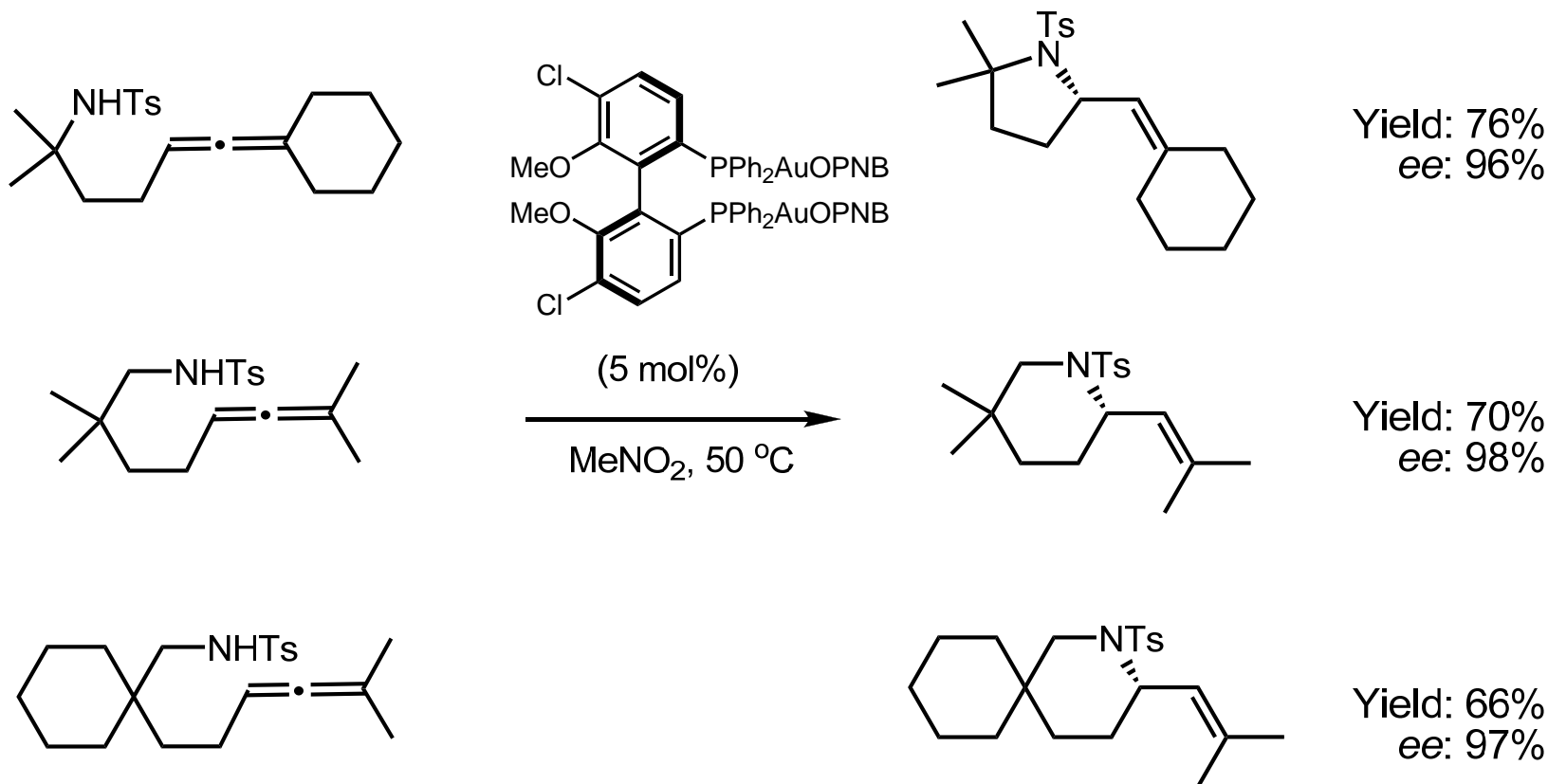
isolated catalyst	17	88	98
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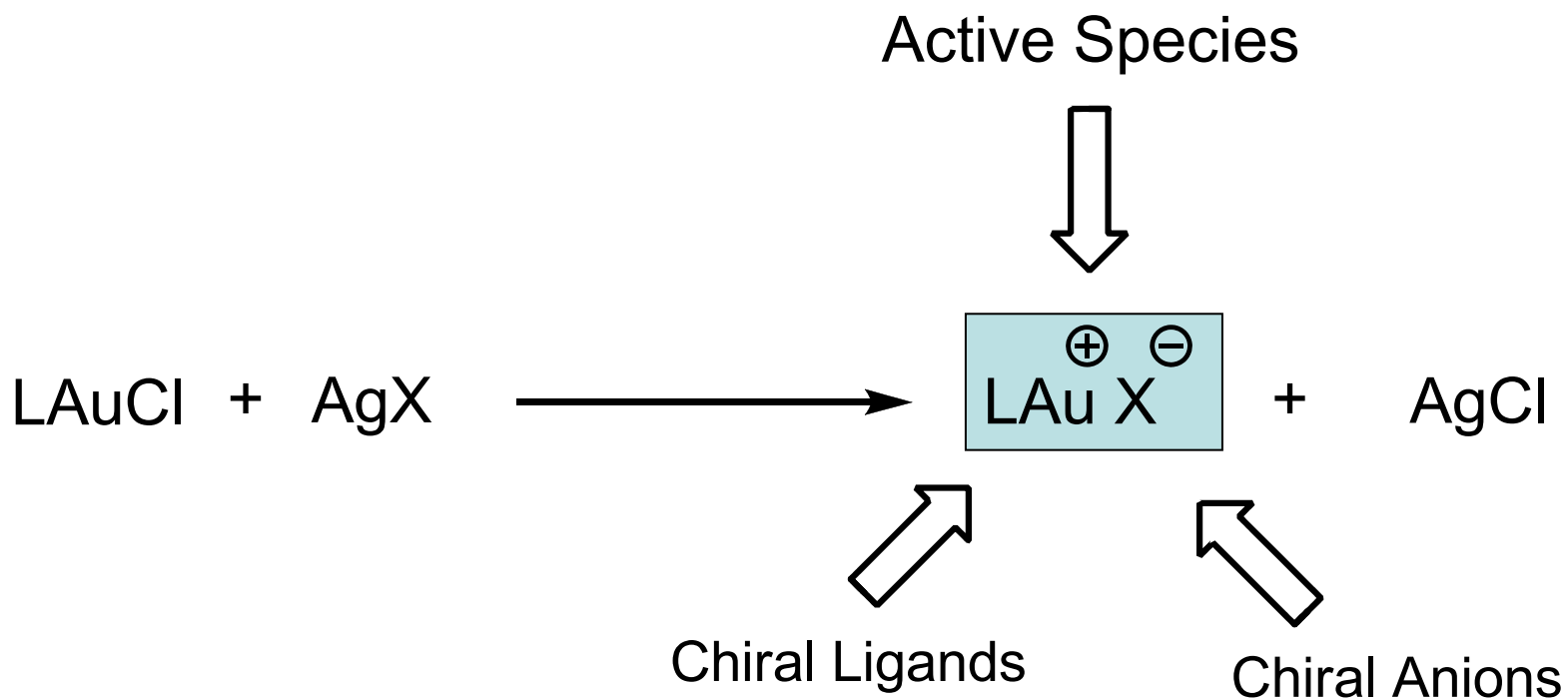
Substrate Scope for Hydroamination



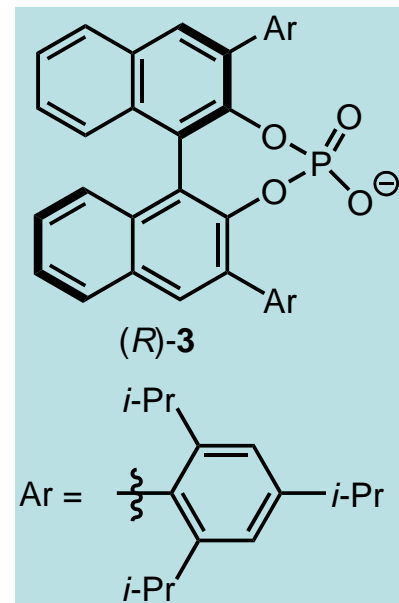
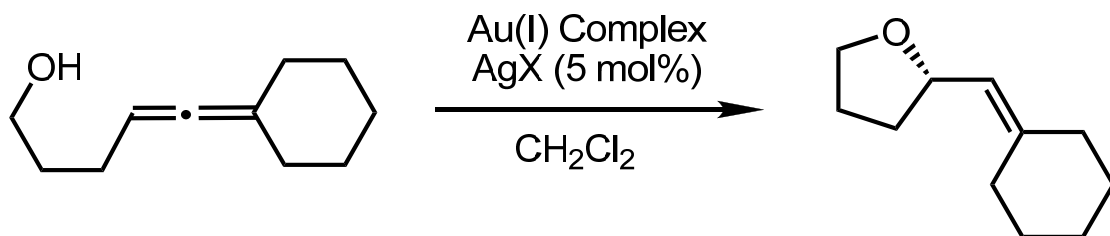
Substrate Scope for Hydroamination



Chirality in Active Species

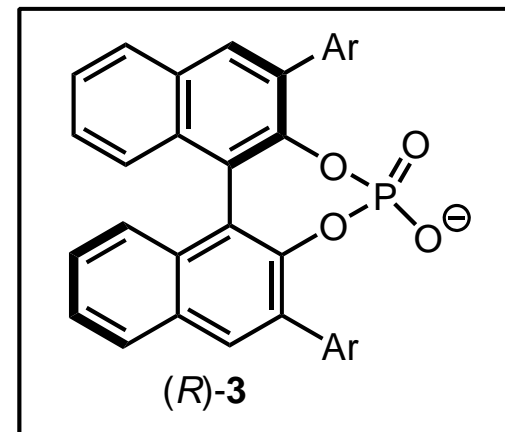
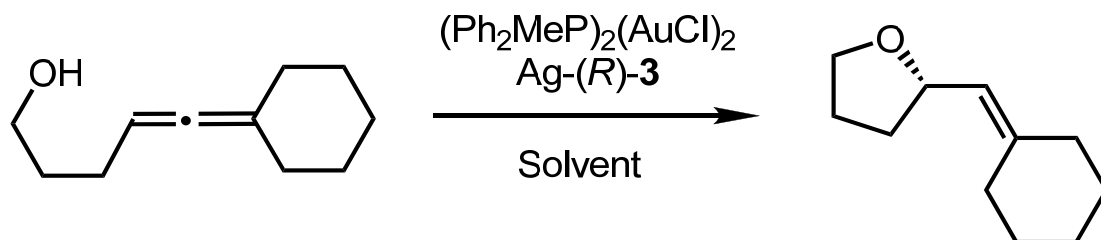


Chiral Anion Directed Hydroalkoxylation



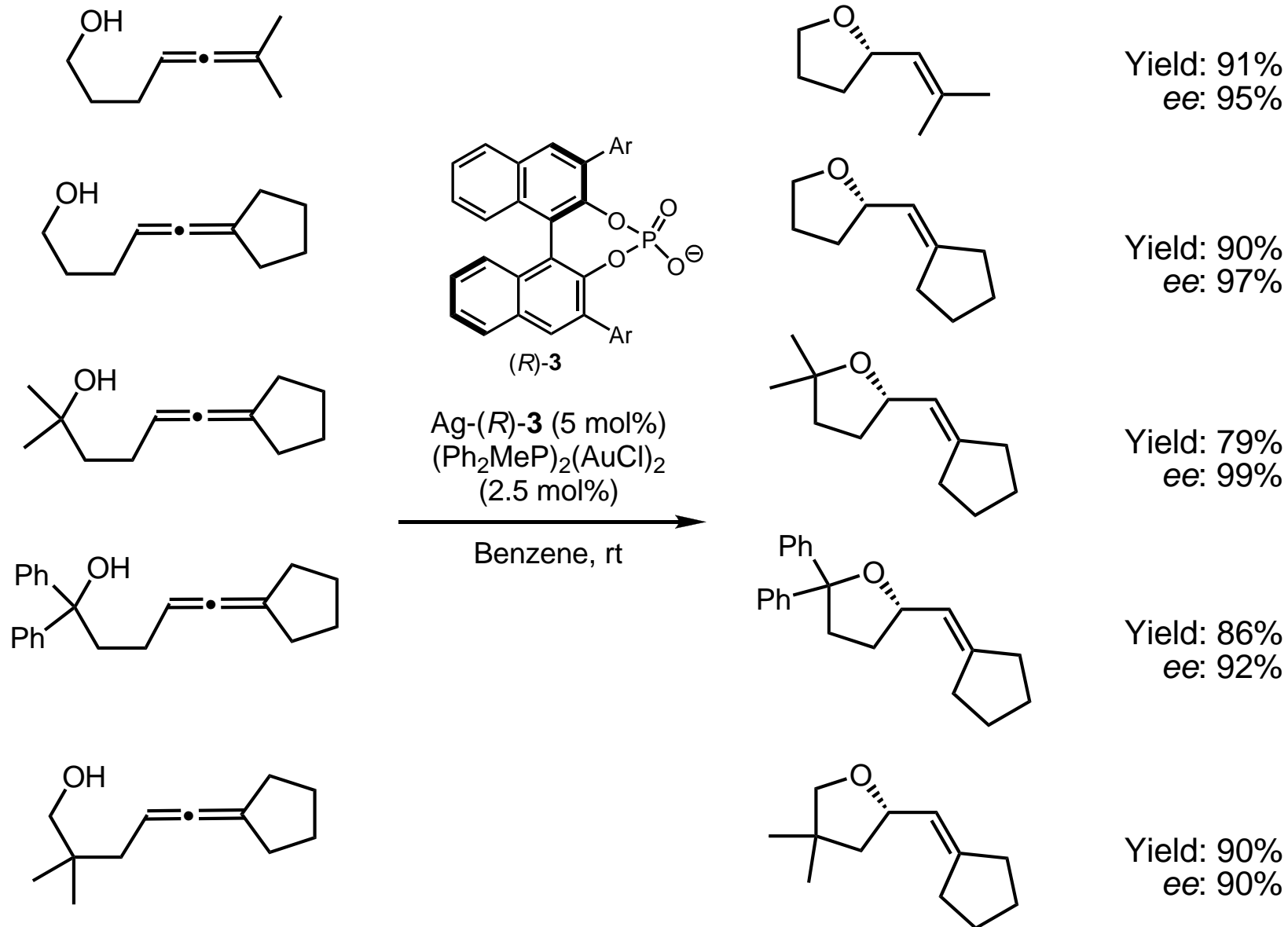
Au(I) Complex	X	Yield (%)	ee (%)
Ph ₃ PAuCl (5 mol%)	(<i>R</i>)-3	89	48
(Ph ₂ MeP) ₂ (AuCl) ₂ (2.5 mol%)	(<i>R</i>)-3	76	65

Screening for Solvents



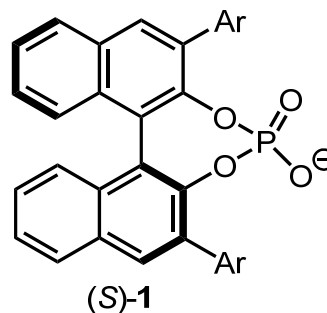
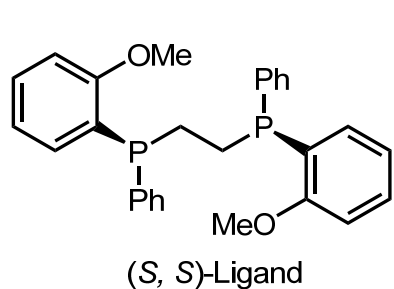
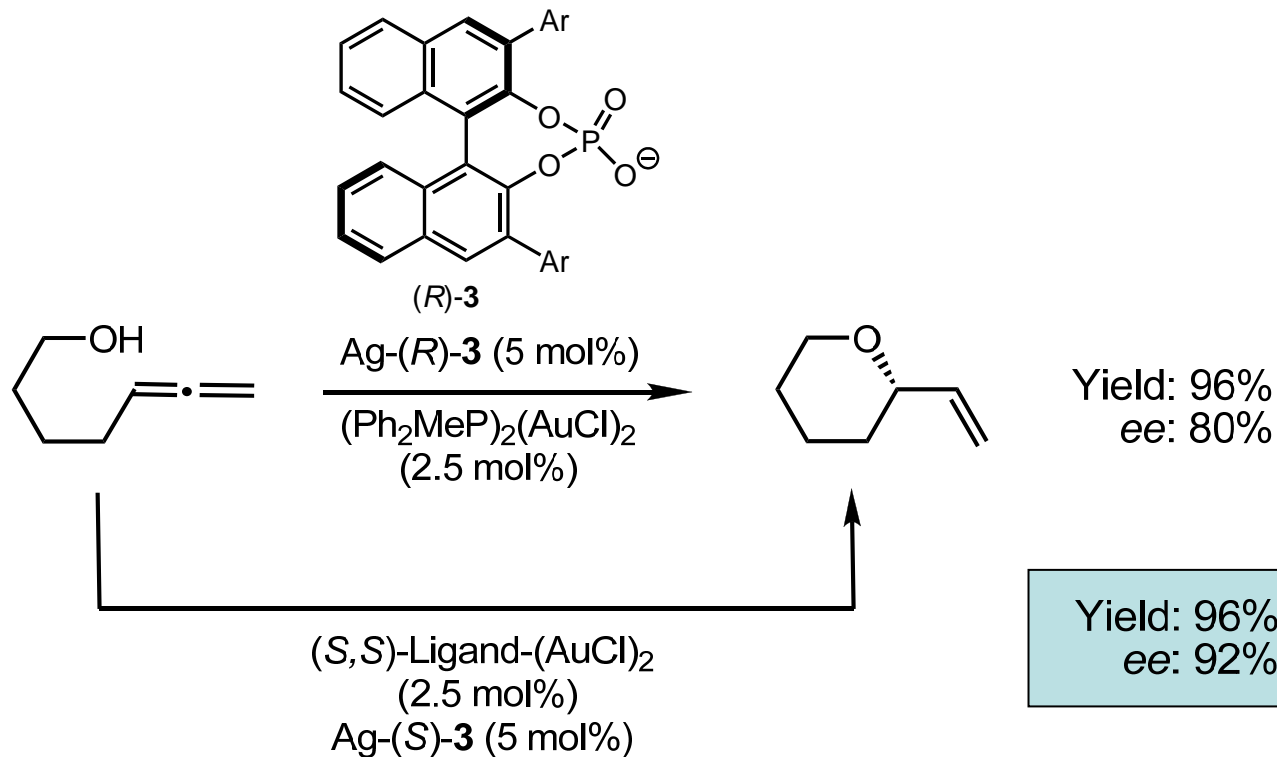
Solvent	Yield (%)	ee (%)
Nitromethane	60	18
Acetone	71	37
Dichloromethane	76	65
Tetrahydrofuran	83	76
Benzene	90	97

Substrate Scope for Hydroalkoxylation

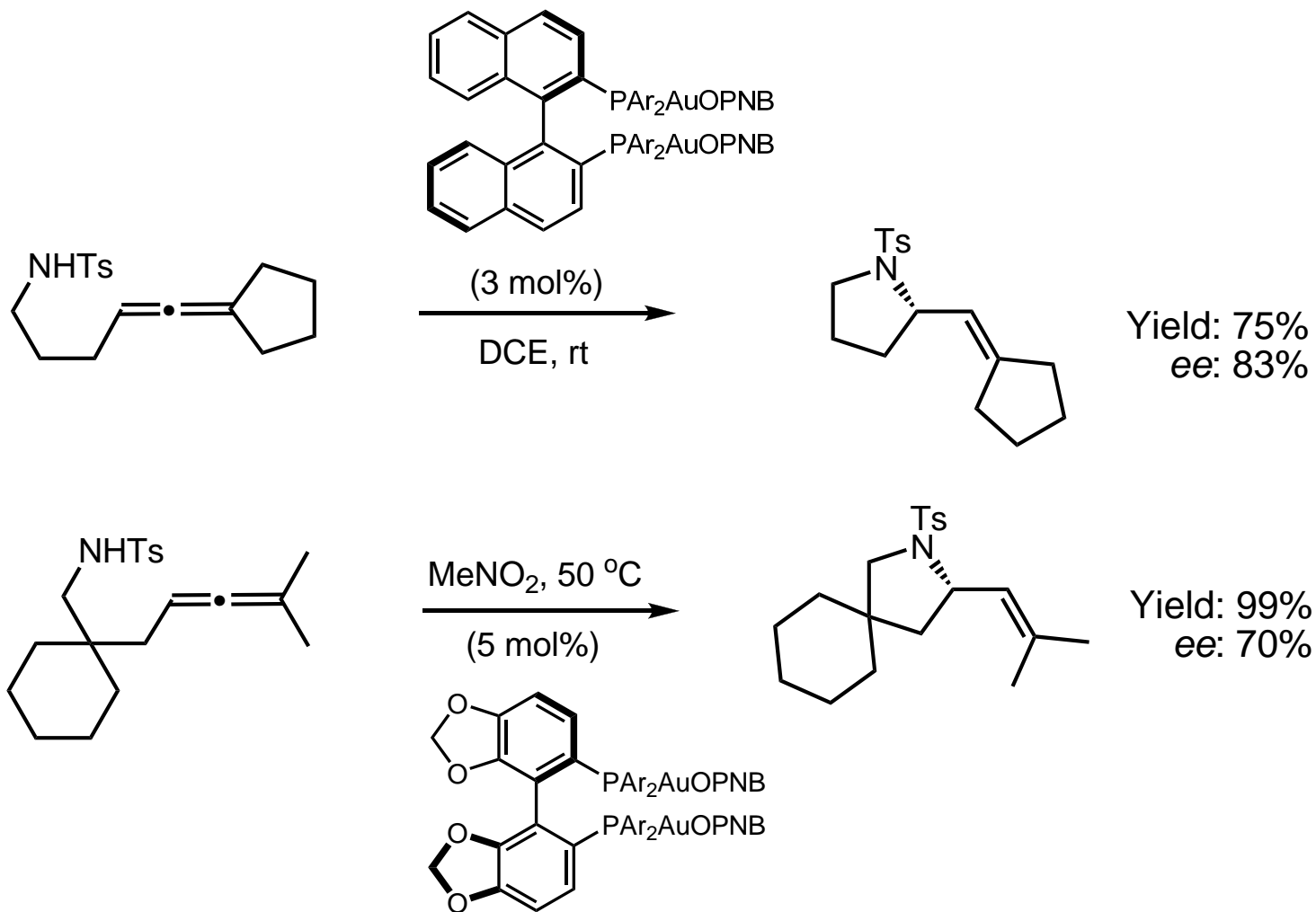


Hamilton, G. L.; Kang, E. J.; Mba, M.; Toste, F. D. *Science*, **2007**, *317*, 496.

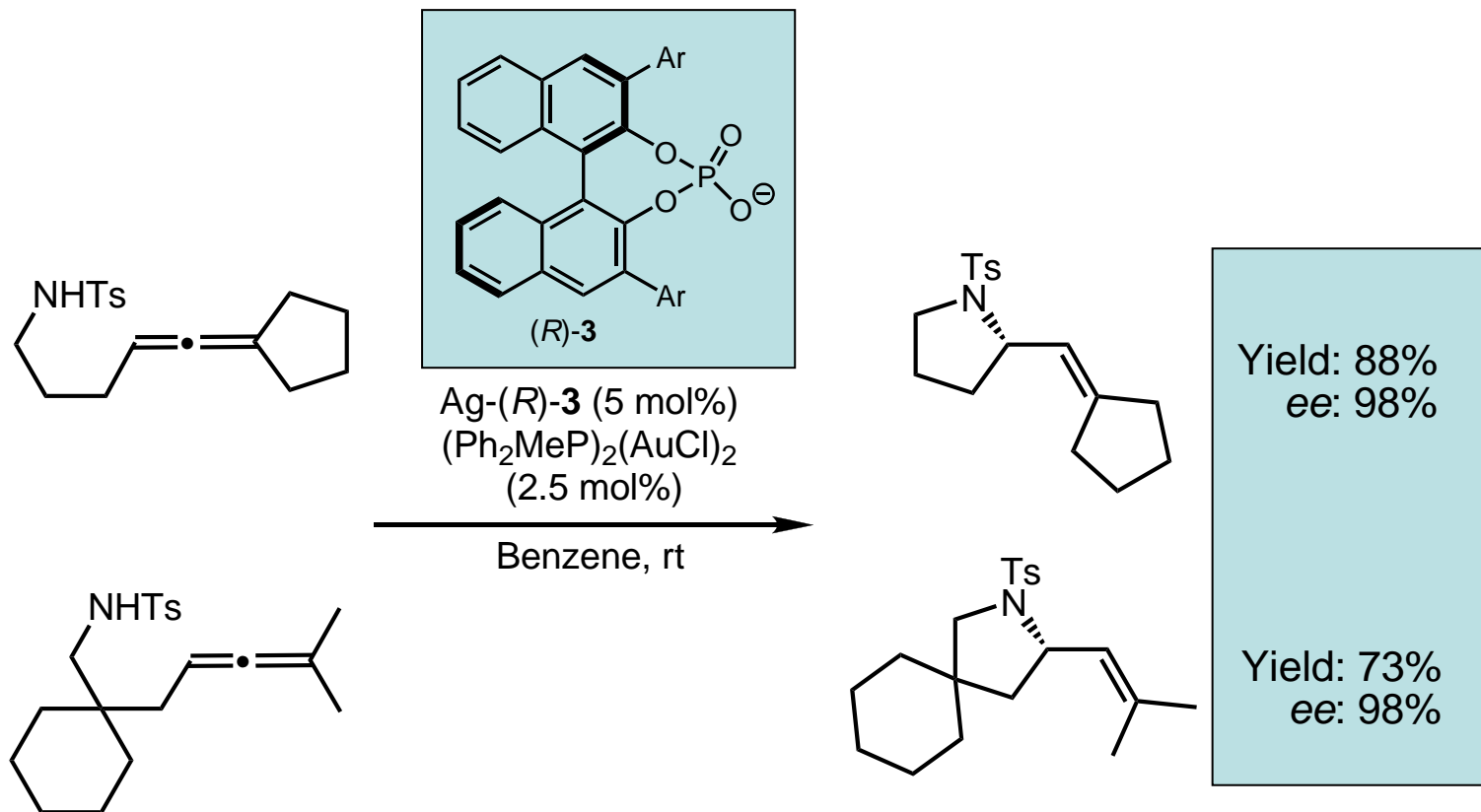
Matched Case for Hydroalkoxylation



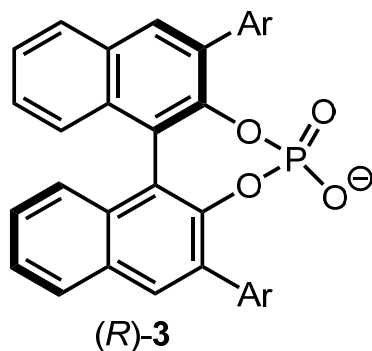
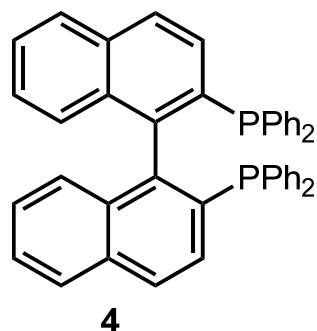
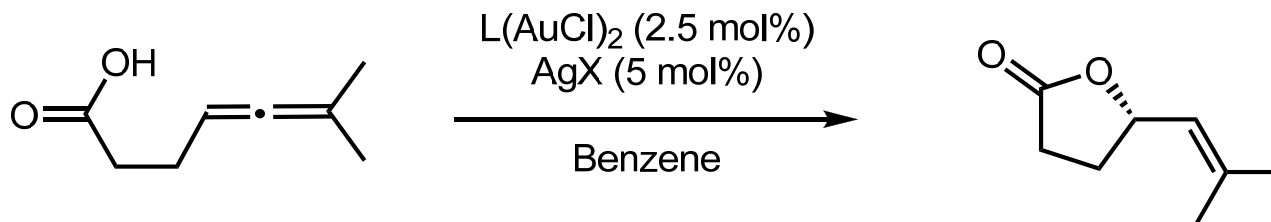
Hydroamination Using Chiral Ligands



Hydroamination Using Chiral Counterion



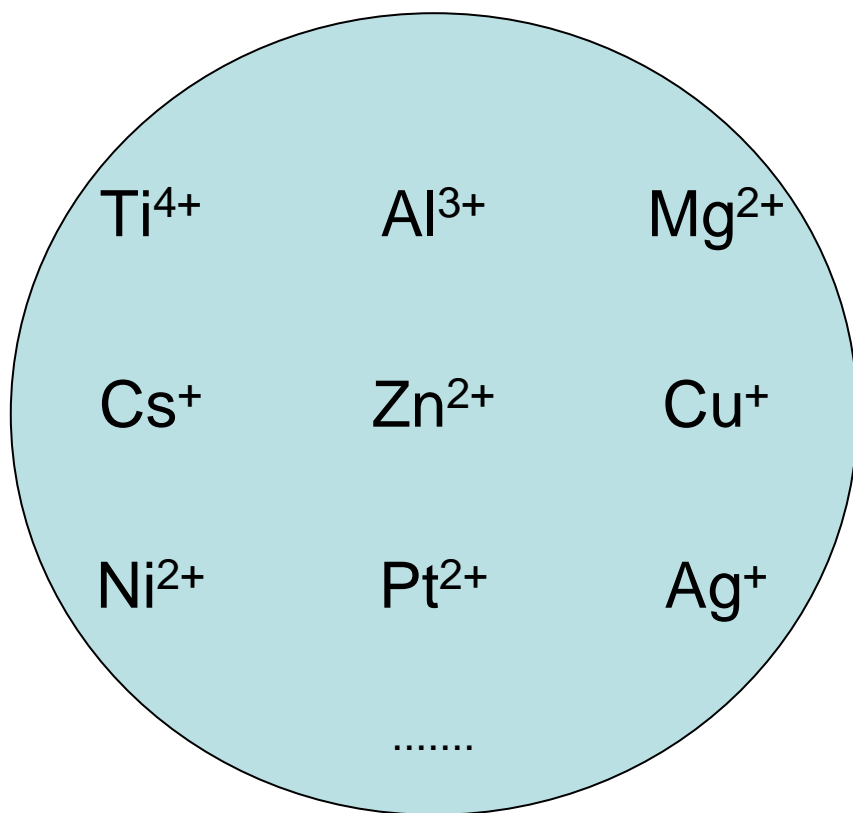
Matched and Mismatched Case in Hydrocarboxylation



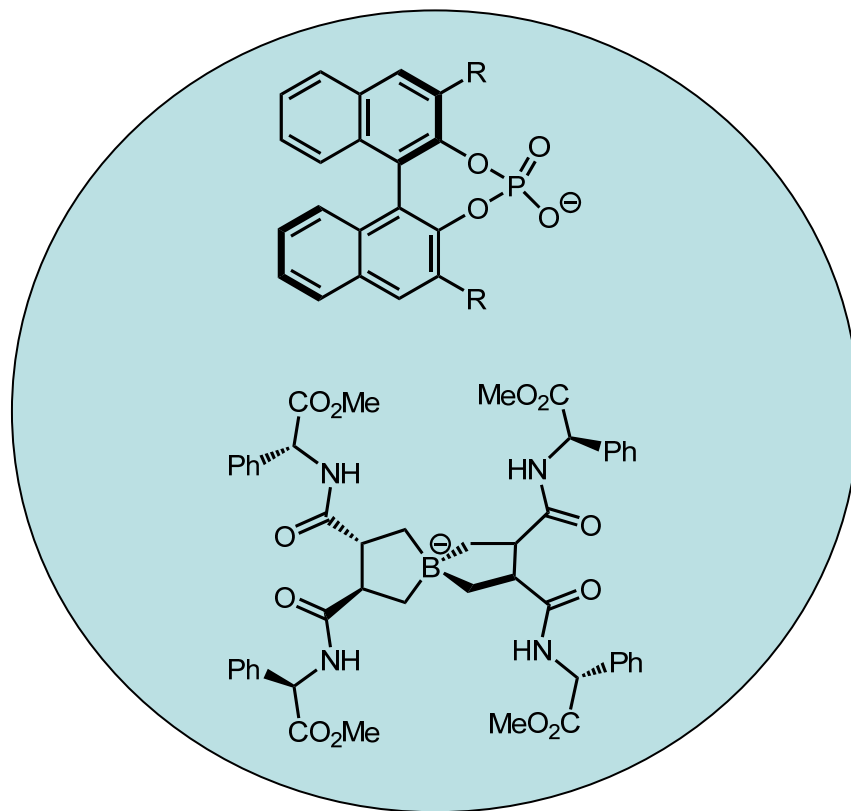
L	X	Yield (%)	ee (%)
(<i>R</i>)- 4	4-(NO_2)- C_6H_3 - COO^-	80	-38
(Ph_2MeP) ₂	(<i>R</i>)- 3	89	12
(<i>S</i>)- 4	(<i>R</i>)- 3	88	82
(<i>R</i>)- 4	(<i>R</i>)- 3	91	-3

Potentials of Chiral Counterion Strategy

Metal Cationic Catalyst



Chiral Anions



Conclusions

- Gold(I) catalysts are superb Lewis acids for activation of alkynes.
- Several enantioselective reactions catalyzed by gold(I) catalysts have been reported.

In the future,

- More enantioselective reactions could be found to be catalyzed by gold(I) species.
- Work should be done in understanding the mechanism of the enantioselective reaction.

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Zhensheng, Nilanjina, Alex, Kostas, Victor